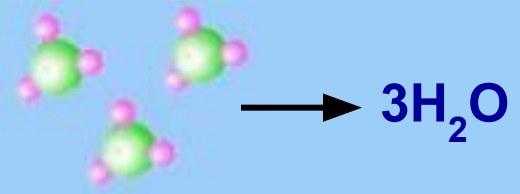
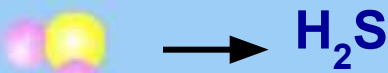
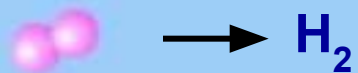
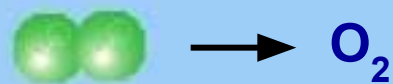


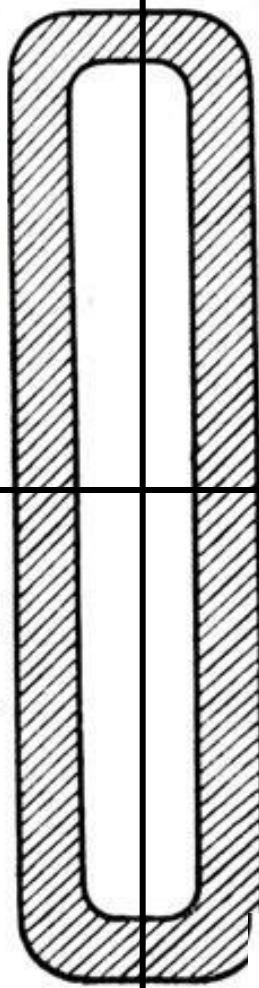
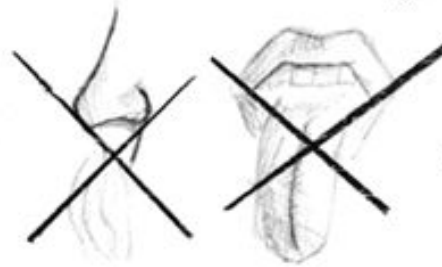
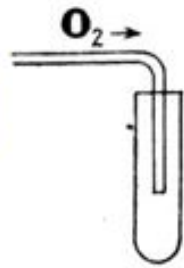
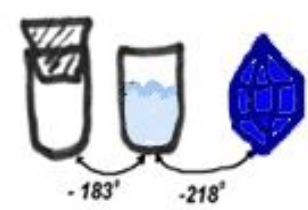
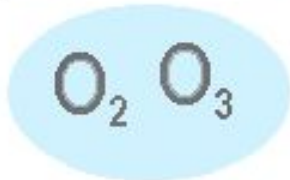
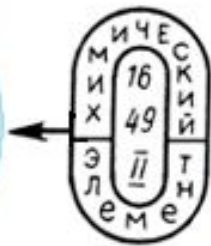
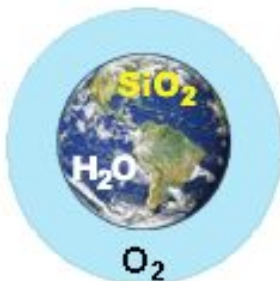
$14\ 14 \rightarrow \text{Mr}(\text{N}_2) = 14 \times 2 = 28$

$1\ 1\ 16 \rightarrow \text{Mr}(\text{H}_2\text{O}) = 1 \times 2 + 1 \times 16 = 18$

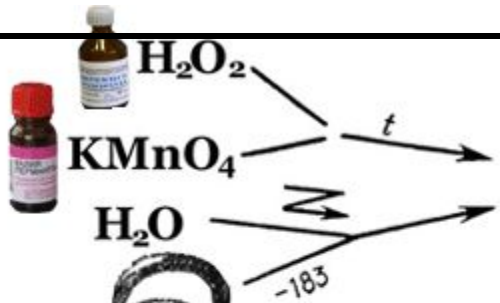
$16\ 32\ 16\ 16 \rightarrow \text{Mr}(\text{SO}_3) = 1 \times 32 + 3 \times 16 = 80$

$1 \rightarrow \text{Ar}(\text{H}) = 1$
 $16 \rightarrow \text{Ar}(\text{O}) = 16$
 $32 \rightarrow \text{Ar}(\text{S}) = 32$

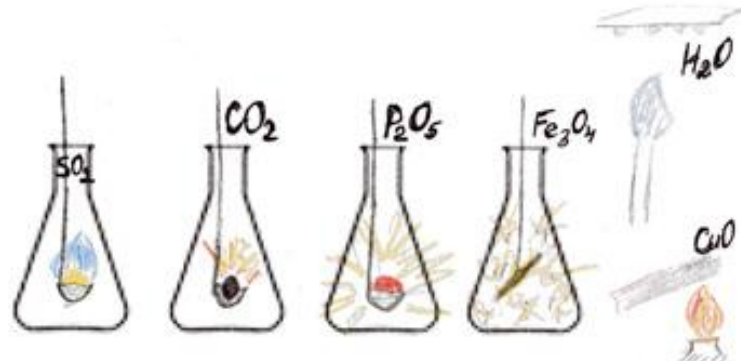
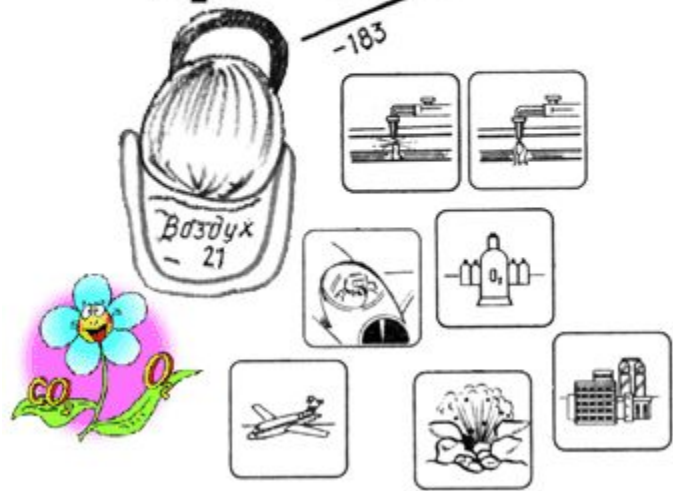
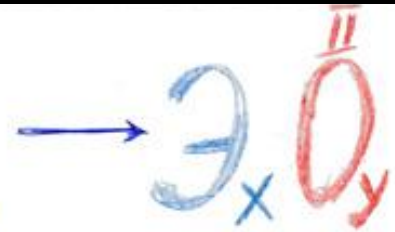




2



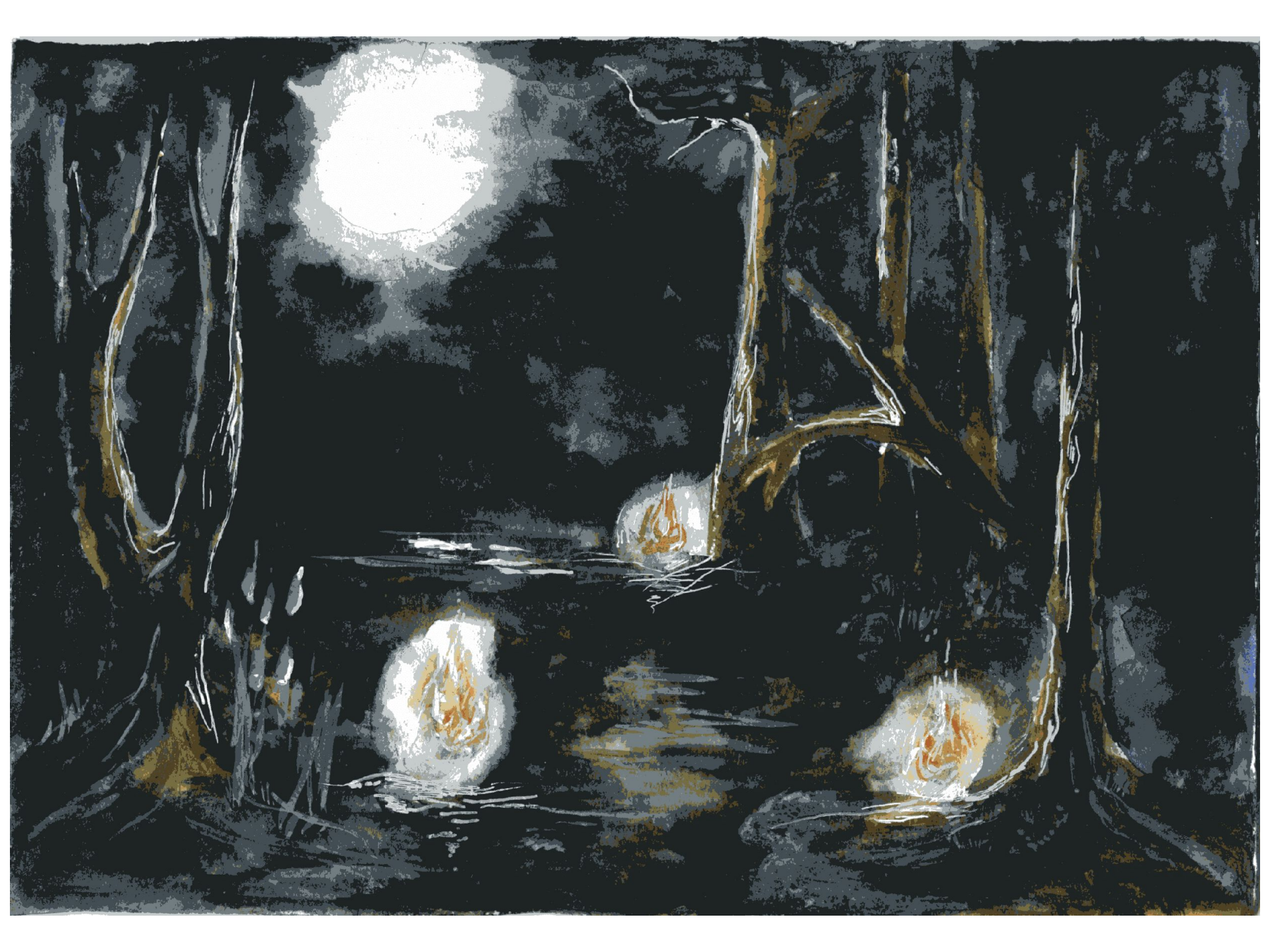
+ Me
не Me
СЛОЖНЫЕ
Вещества



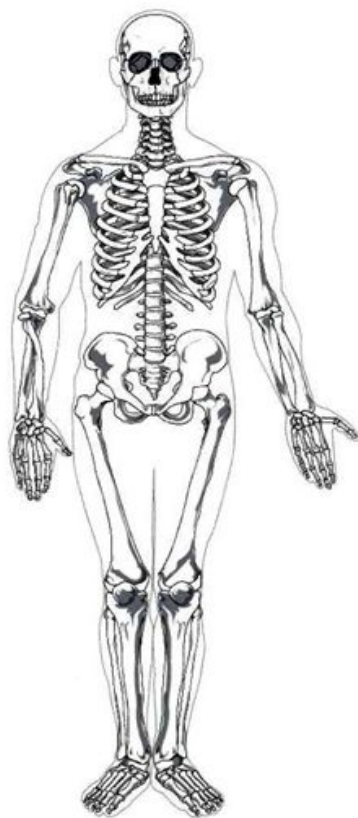








ФОСФОР В ОРГАНИЗМЕ ЧЕЛОВЕКА



1,4 кг



130 г



12 г

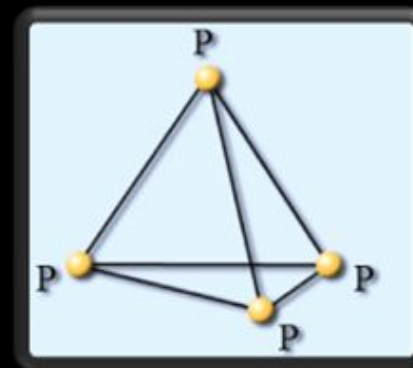
Белый фосфор (P_4)

■ $t_{\text{плав.}} = 44,1^{\circ}\text{C}$

■ $t_{\text{кип.}} = 280^{\circ}\text{C}$

■ $P_4 \rightarrow 2P_2$ (800°C)

■ α -форма ($1,823 \text{ г/см}^3$) \rightarrow β -форма ($1,88 \text{ г/см}^3$)



$-76,9^{\circ}\text{C}$



0,05 – 0,1 Г.

1%

5%



