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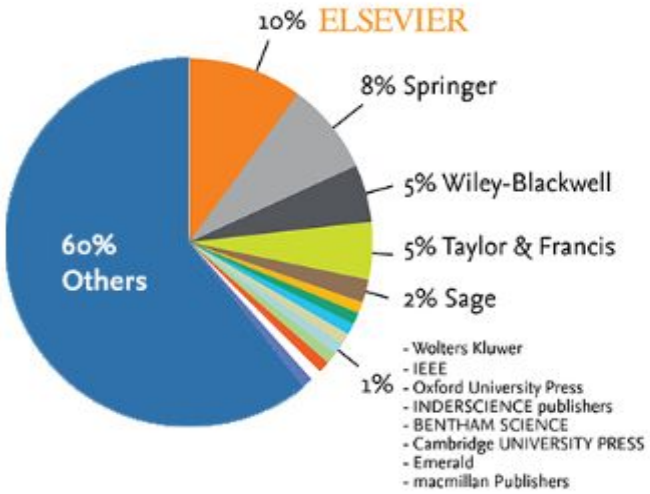

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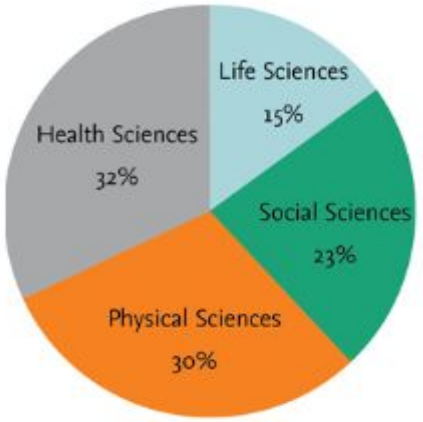

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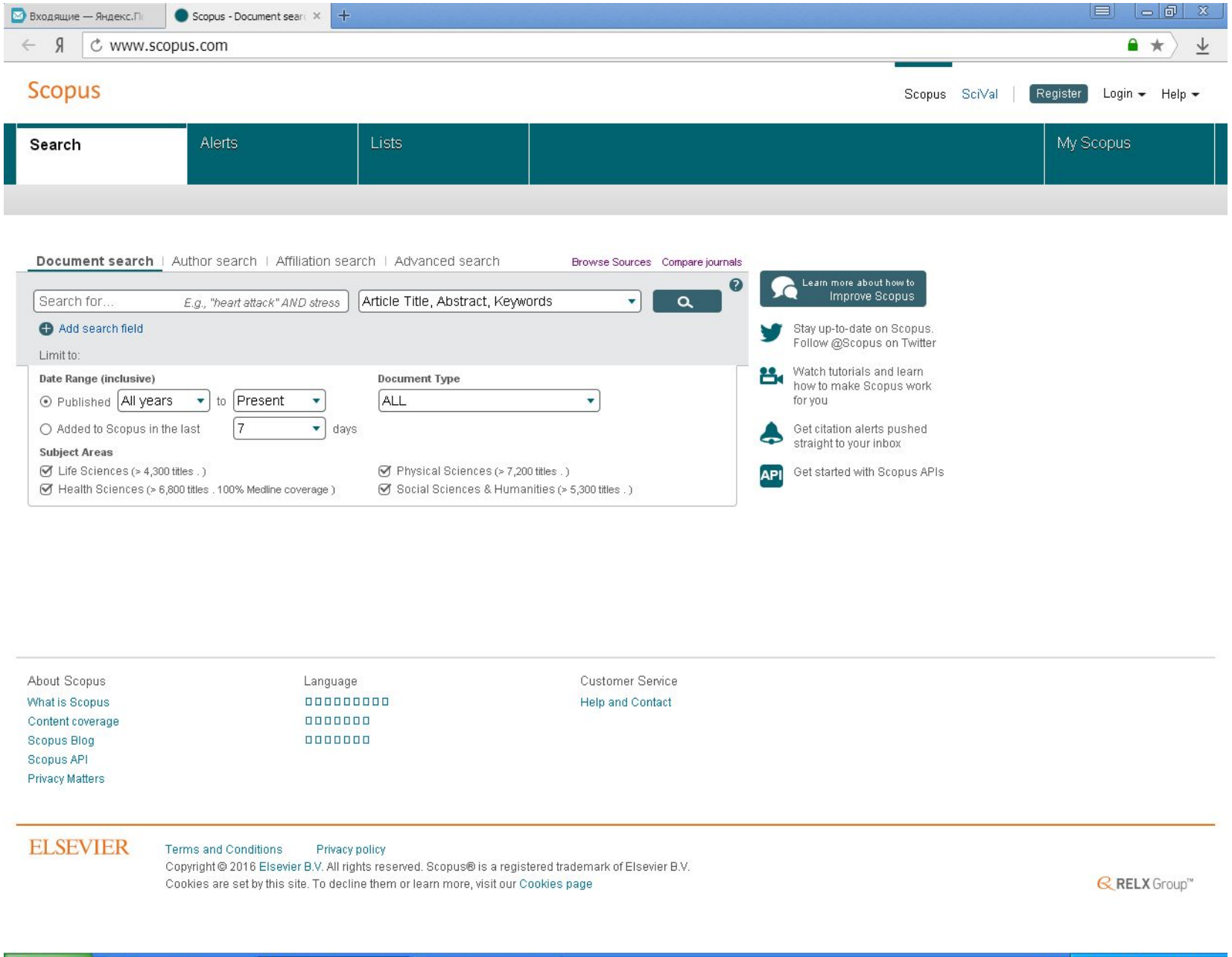

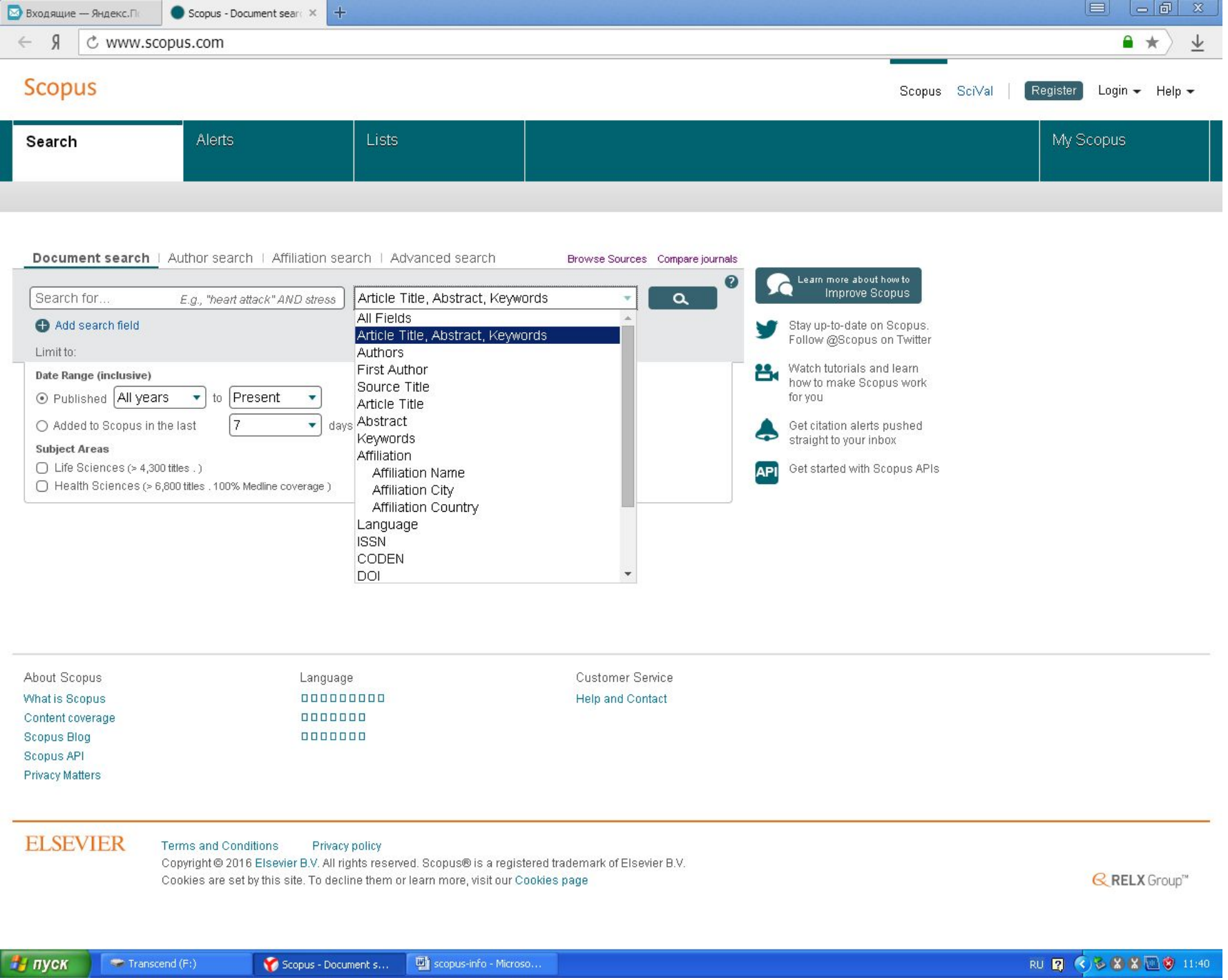
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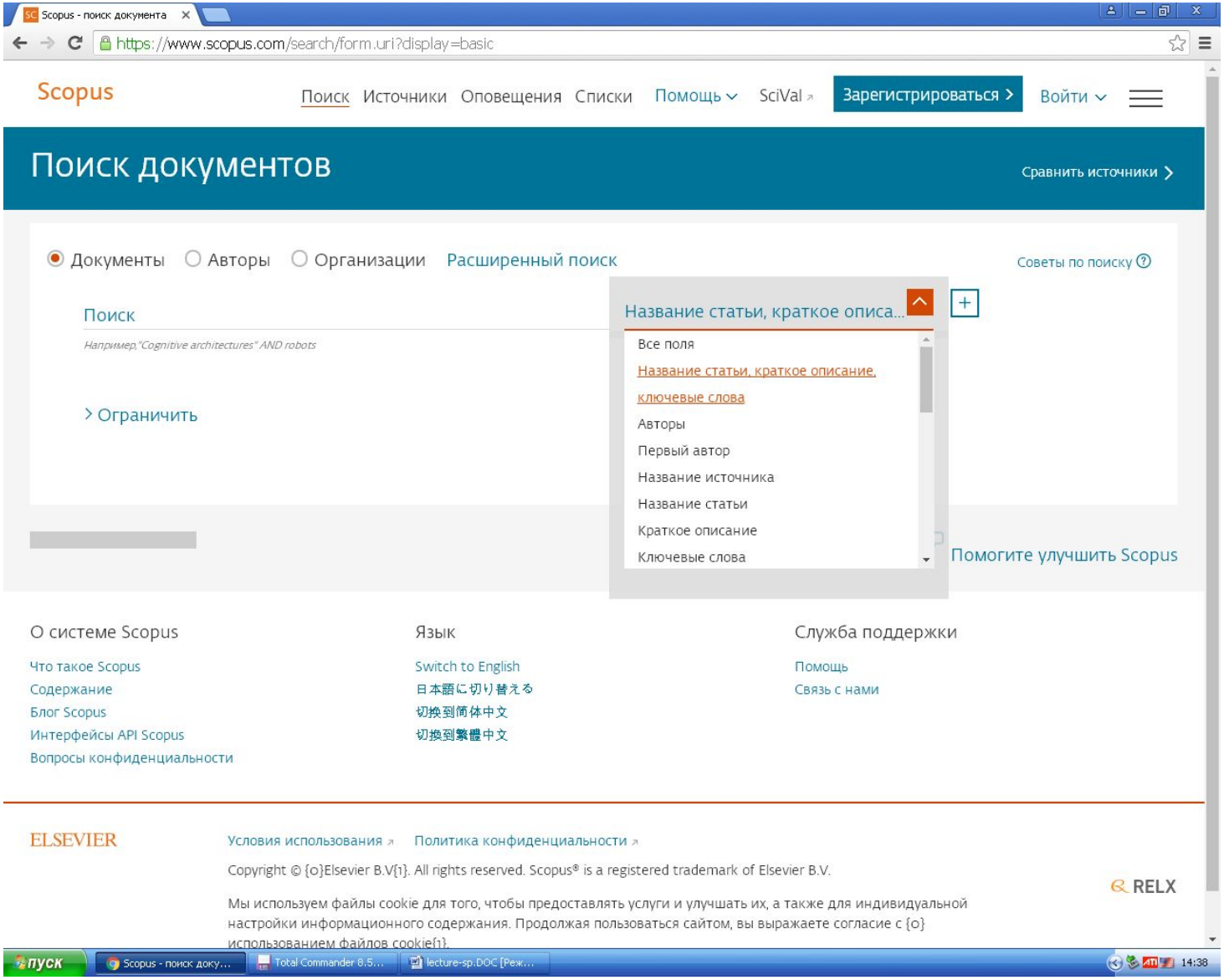
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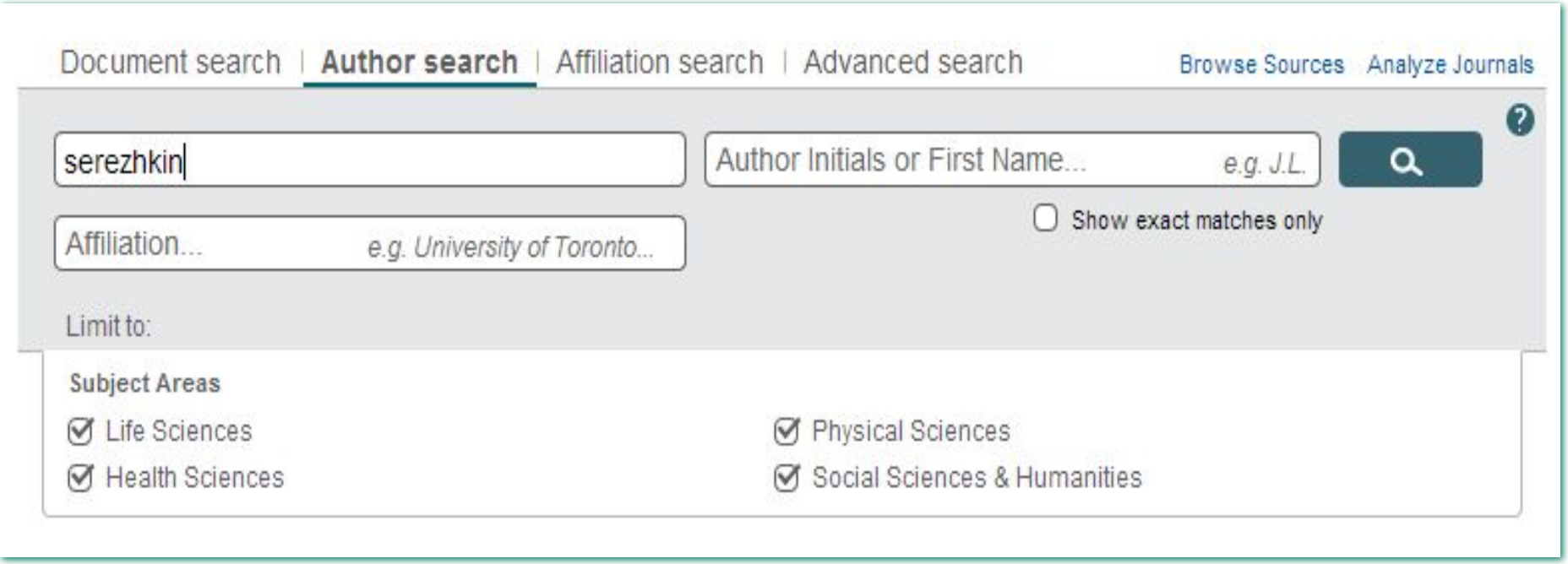

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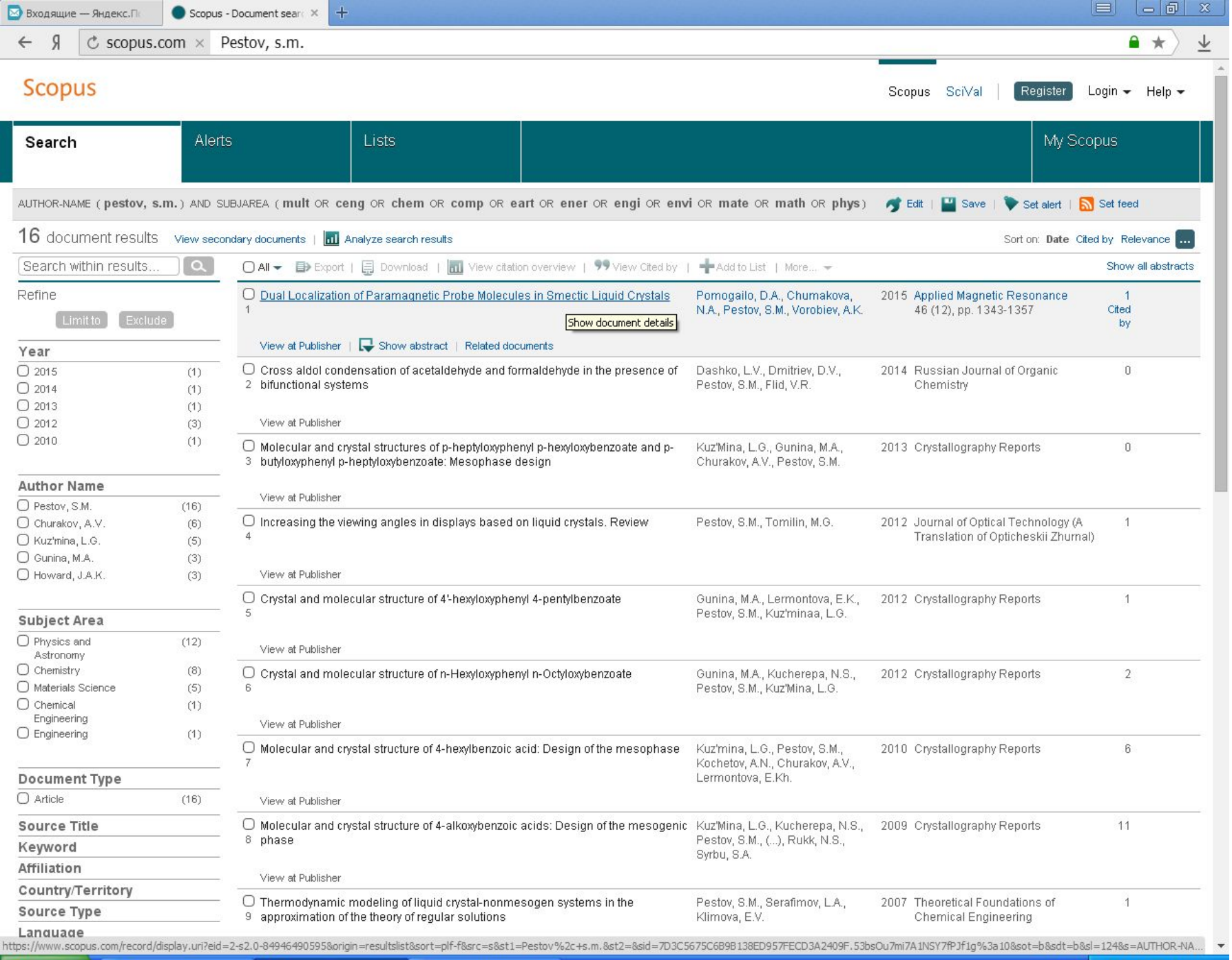
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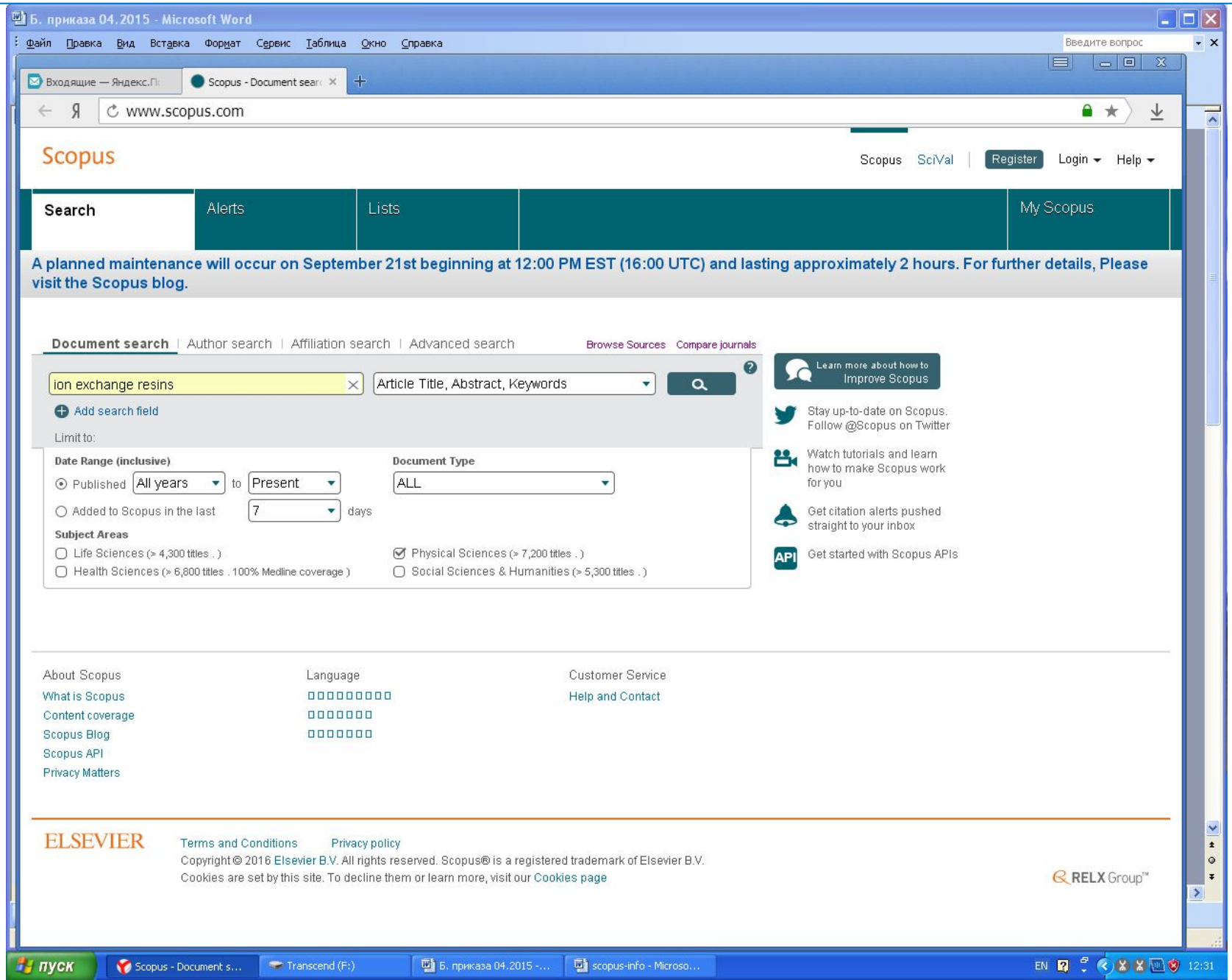

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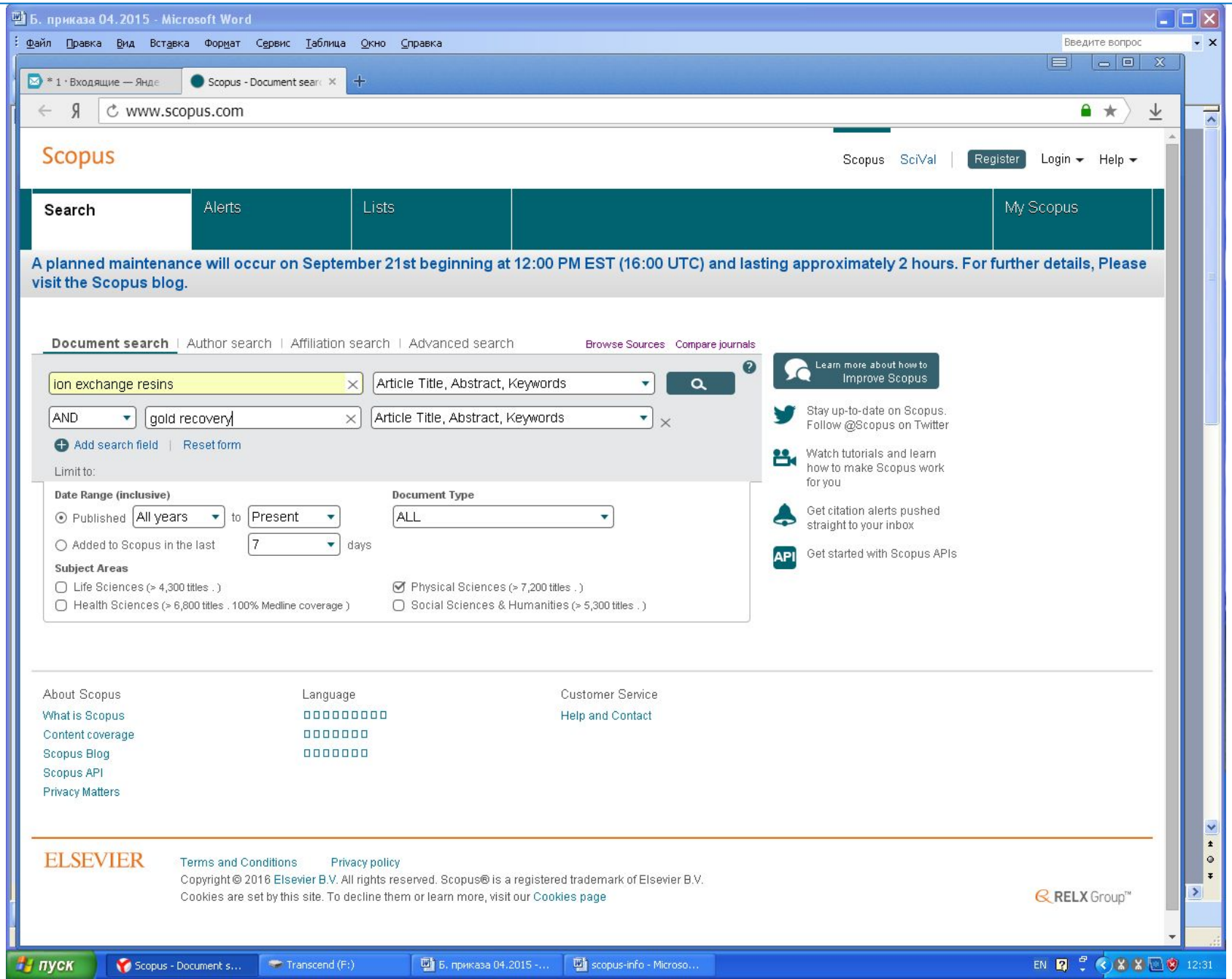

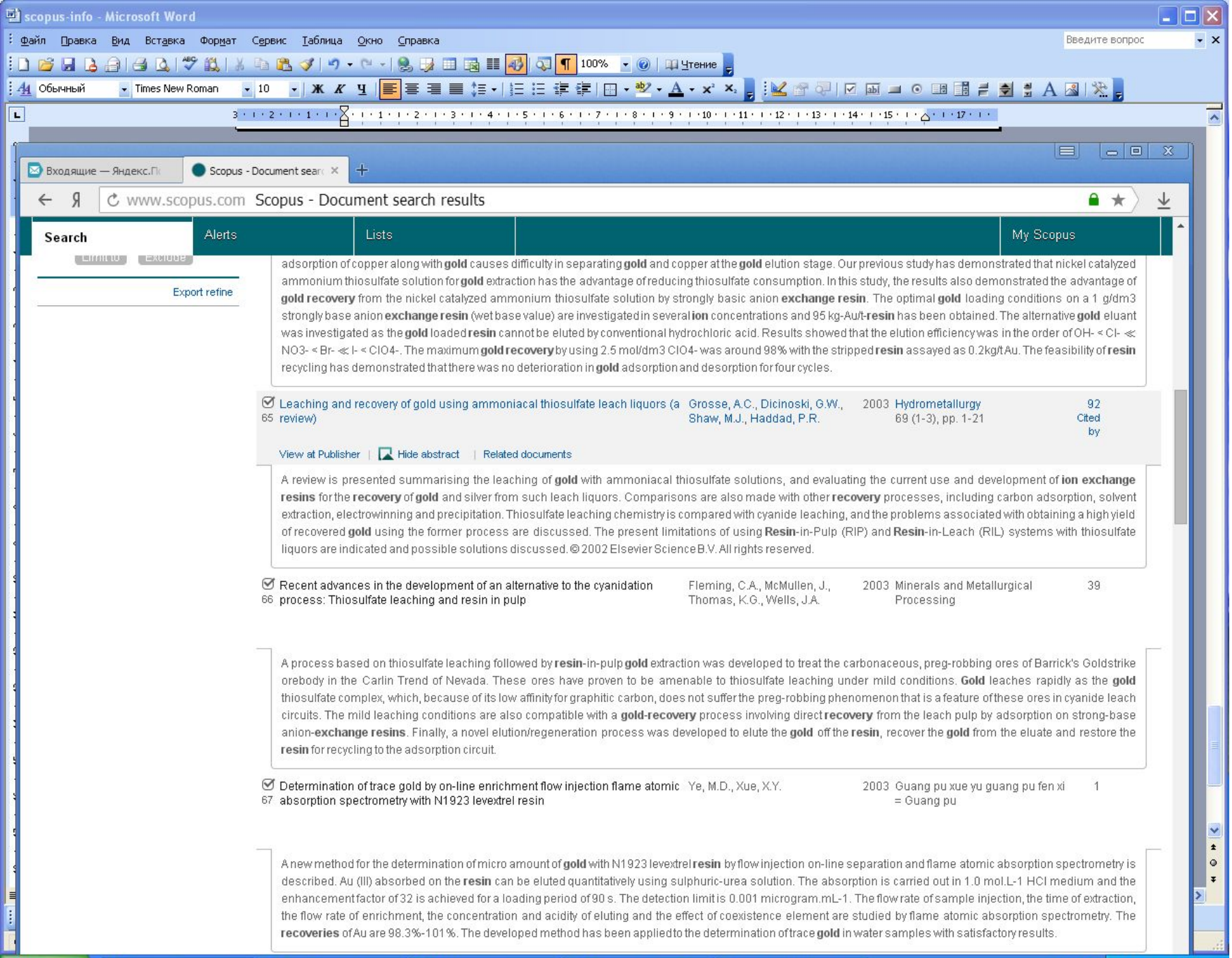
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adsorption of copper along with gold causes difficulty in separating gold and copper at the gold elution stage. Our previous study has demonstrated that nickel catalyzed ammonium thiosulfate solution for gold extraction has the advantage of reducing thiosulfate consumption. In this study, the results also demonstrated the advantage of gold recovery from the nickel catalyzed ammonium thiosulfate solution by strongly basic anion exchange resin. The optimal gold loading conditions on a 1 g/dm3 strongly base anion exchange resin (wet base value) are investigated in several ion concentrations and 95 kg-Au/t-resin has been obtained. The alternative gold eluant was investigated as the gold loaded resin cannot be eluted by conventional hydrochloric acid. Results showed that the elution efficiency was in the order of OH- < Cl- ≪ NO3- < Br- ≪ I- < ClO4-. The maximum gold recovery by using 2.5 mol/dm3 ClO4- was around 98% with the stripped resin assayed as 0.2kg/t Au. The feasibility of resin recycling has demonstrated that there was no deterioration in gold adsorption and desorption for four cycles.
65 Leaching and recovery of gold using ammoniacal thiosulfate leach liquors (a review)
Grosse, A.C., Dicinoski, G.W., Shaw, M.J., Haddad, P.R.
2003 Hydrometallurgy 69 (1-3), pp. 1-21
92 Cited by
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A review is presented summarising the leaching of gold with ammoniacal thiosulfate solutions, and evaluating the current use and development of ion exchange resins for the recovery of gold and silver from such leach liquors. Comparisons are also made with other recovery processes, including carbon adsorption, solvent extraction, electrowinning and precipitation. Thiosulfate leaching chemistry is compared with cyanide leaching, and the problems associated with obtaining a high yield of recovered gold using the former process are discussed. The present limitations of using Resin-in-Pulp (RIP) and Resin-in-Leach (RIL) systems with thiosulfate liquors are indicated and possible solutions discussed. © 2002 Elsevier Science B.V. All rights reserved.
66 Recent advances in the development of an alternative to the cyanidation process: Thiosulfate leaching and resin in pulp
Fleming, C.A., McMullen, J., Thomas, K.G., Wells, J.A.
2003 Minerals and Metallurgical Processing
39
A process based on thiosulfate leaching followed by resin-in-pulp gold extraction was developed to treat the carbonaceous, preg-robbing ores of Barrick's Goldstrike orebody in the Carlin Trend of Nevada. These ores have proven to be amenable to thiosulfate leaching under mild conditions. Gold leaches rapidly as the gold thiosulfate complex, which, because of its low affinity for graphitic carbon, does not suffer the preg-robbing phenomenon that is a feature of these ores in cyanide leach circuits. The mild leaching conditions are also compatible with a gold-recovery process involving direct recovery from the leach pulp by adsorption on strong-base anion-exchange resins. Finally, a novel elution/regeneration process was developed to elute the gold off the resin, recover the gold from the eluate and restore the resin for recycling to the adsorption circuit.
67 Determination of trace gold by on-line enrichment flow injection flame atomic absorption spectrometry with N1923 levextrel resin
Ye, M.D., Xue, X.Y.
2003 Guang pu xue yu guang pu fen xi = Guang pu
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A new method for the determination of micro amount of gold with N1923 levextrel resin by flow injection on-line separation and flame atomic absorption spectrometry is described. Au (III) absorbed on the resin can be eluted quantitatively using sulphuric-urea solution. The absorption is carried out in 1.0 mol.L-1 HCl medium and the enhancement factor of 32 is achieved for a loading period of 90 s. The detection limit is 0.001 microgram.mL-1. The flow rate of sample injection, the time of extraction, the flow rate of enrichment, the concentration and acidity of eluting and the effect of coexistence element are studied by flame atomic absorption spectrometry. The recoveries of Au are 98.3%-101%. The developed method has been applied to the determination of trace gold in water samples with satisfactory results.
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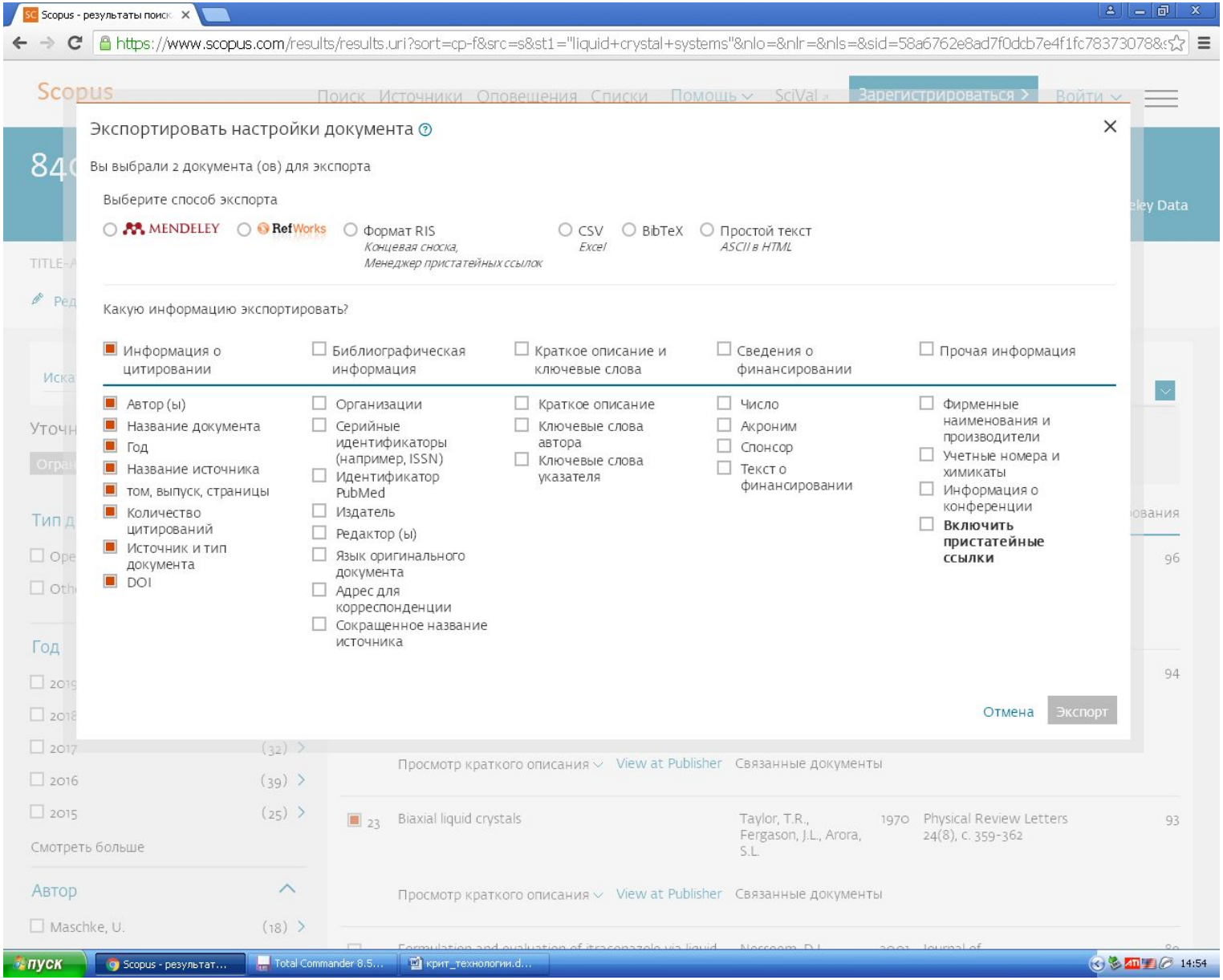

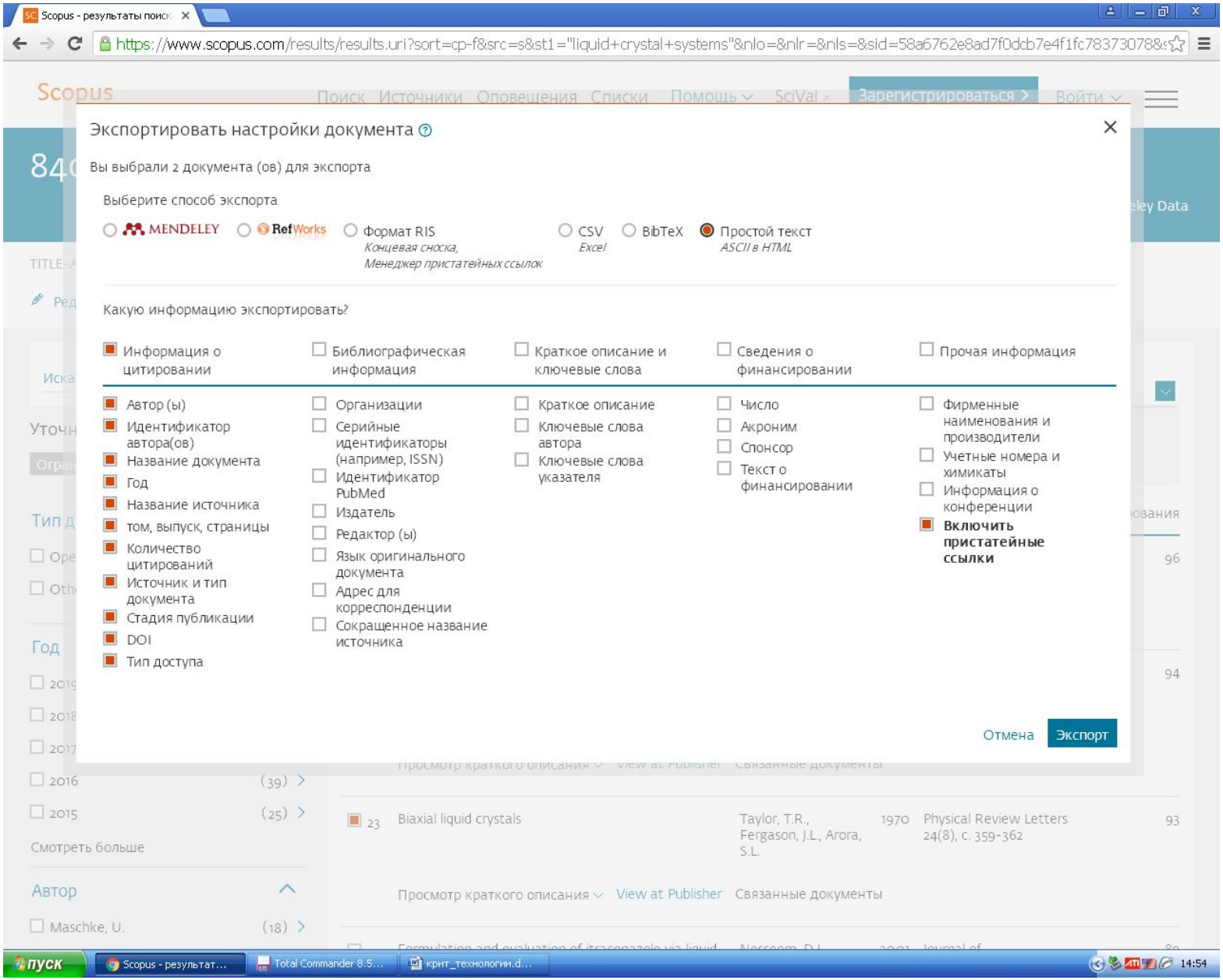
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Mineral Processing and Extractive Metallurgy Review

Volume 37, Issue 2, 3 March 2016, Pages 73-119

Heap leaching technology - Current State, innovations, and future directions: A review (Article)

Ghorbani, Y.[ab], Franzidis, J.-P.[c], Petersen, J.[c]

[a] Camborne School of Mines, College of Engineering, Mathematics and Physical Sciences (CEMPS), University of Exeter, Cornwall, United Kingdom

[b] CICITEM, Centro de Investigación Científico Tecnológico para la Minería, Antofagasta, Chile

[c] Minerals to Metals Signature Theme, Department of Chemical Engineering, University of Cape Town, Private Bag X6, Rondebosch, South Africa

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Abstract

Heap leaching is a well-established extractive metallurgical technology enabling the economical processing of various kinds of low-grade ores, which could not otherwise be exploited. However, despite much progress since it was first applied in recent times, the process remains limited by low recoveries and long extraction times. It is becoming increasingly clear that the choice of heap leaching as a suitable technology to process a particular mineral resource, which is both environmentally sound and economically viable, very much depends on having a comprehensive understanding of the underlying fundamental mechanisms of the processes and how they interact with the particular mineralogy of the ore body under consideration. This paper provides an introduction to the theoretical background of various heap leach processes, offers a scientific and patent literature overview on technology developments in commercial heap leaching operations around the world, identifies factors that drive the selection of heap leaching as a processing technology, describes challenges to exploiting these innovations, and concludes with a discussion on the future of heap leaching. © 2016 Taylor & Francis.

Author keywords

Agglomeration; copper; gold; heap leaching; hydrometallurgy; mineralogy

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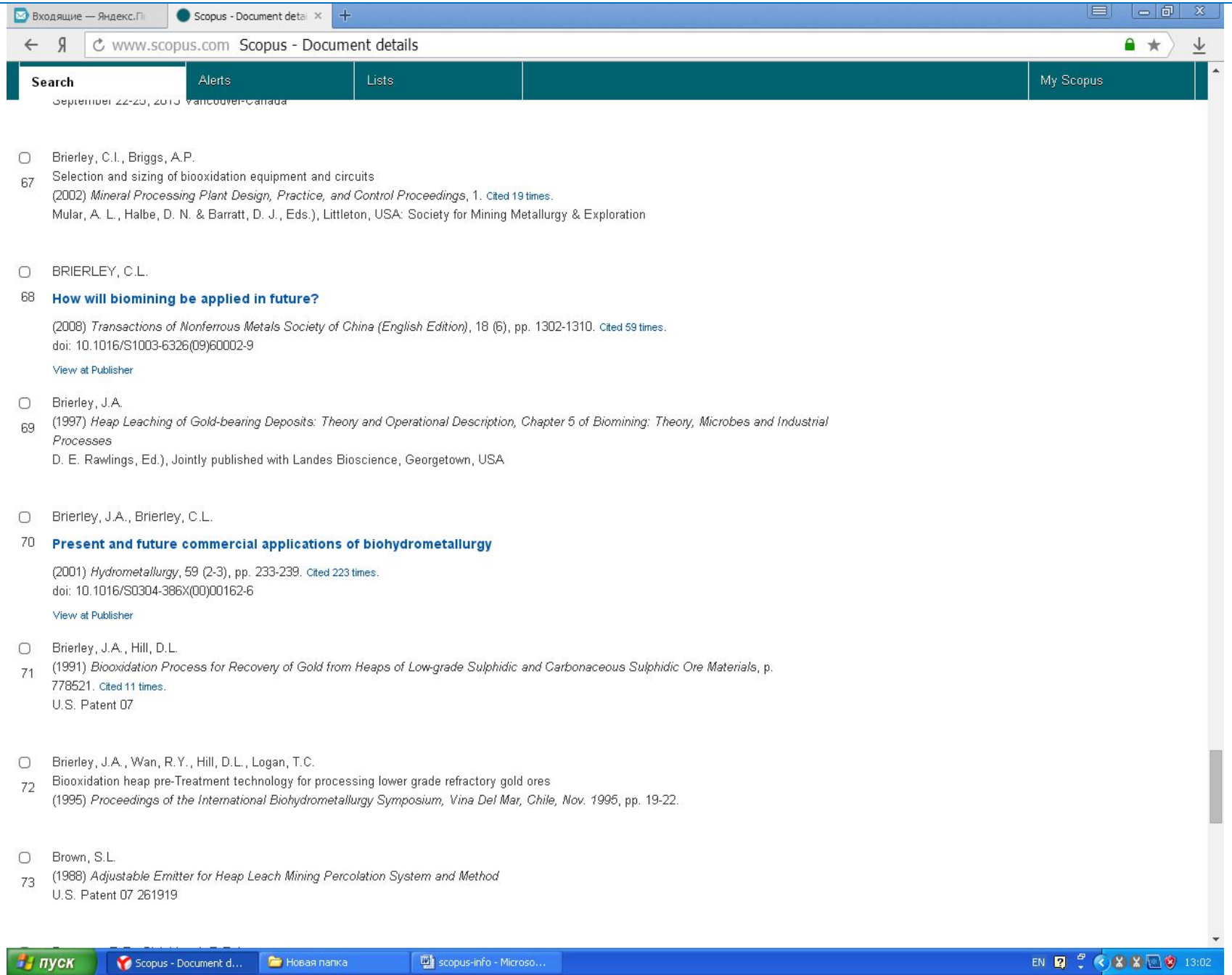

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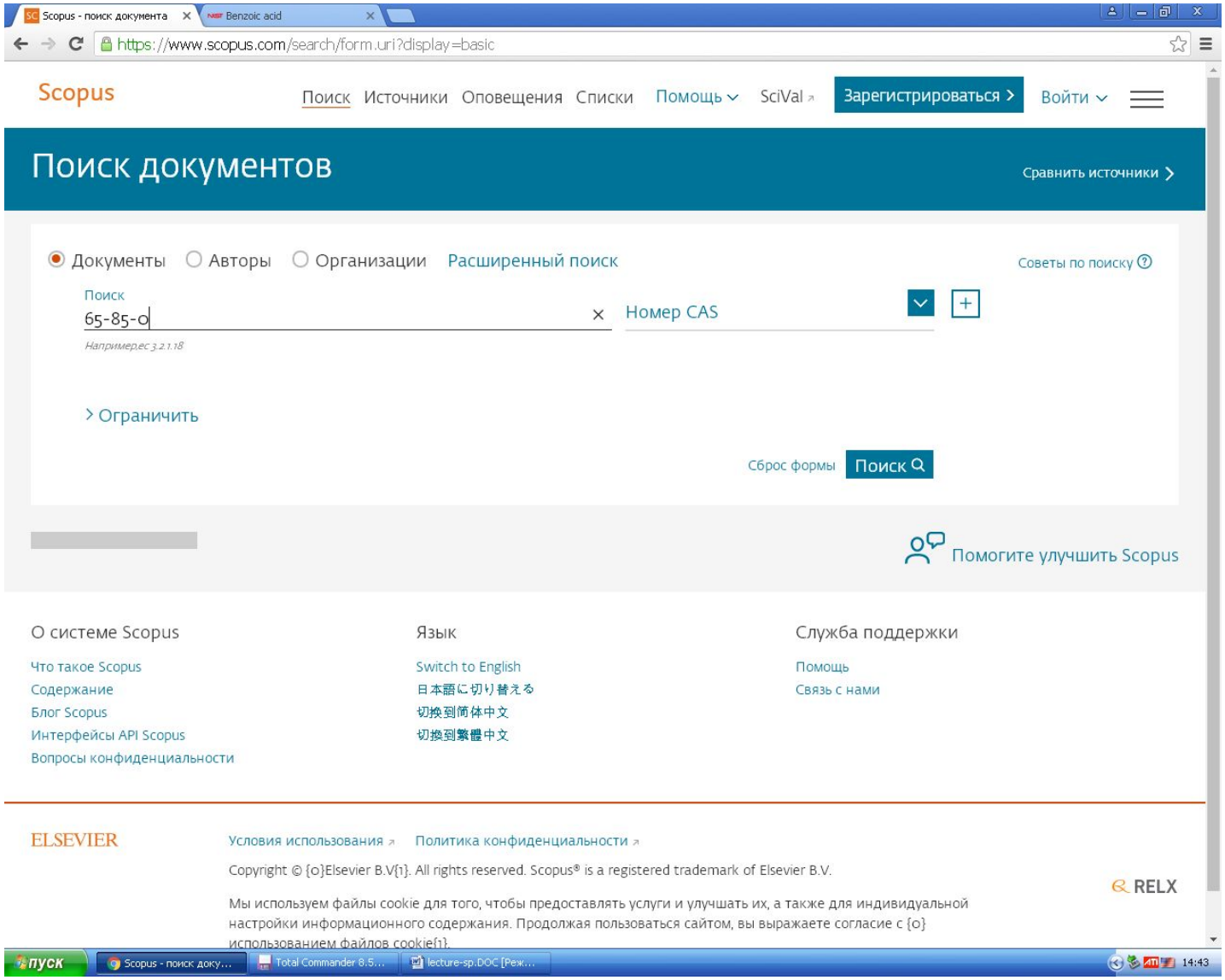

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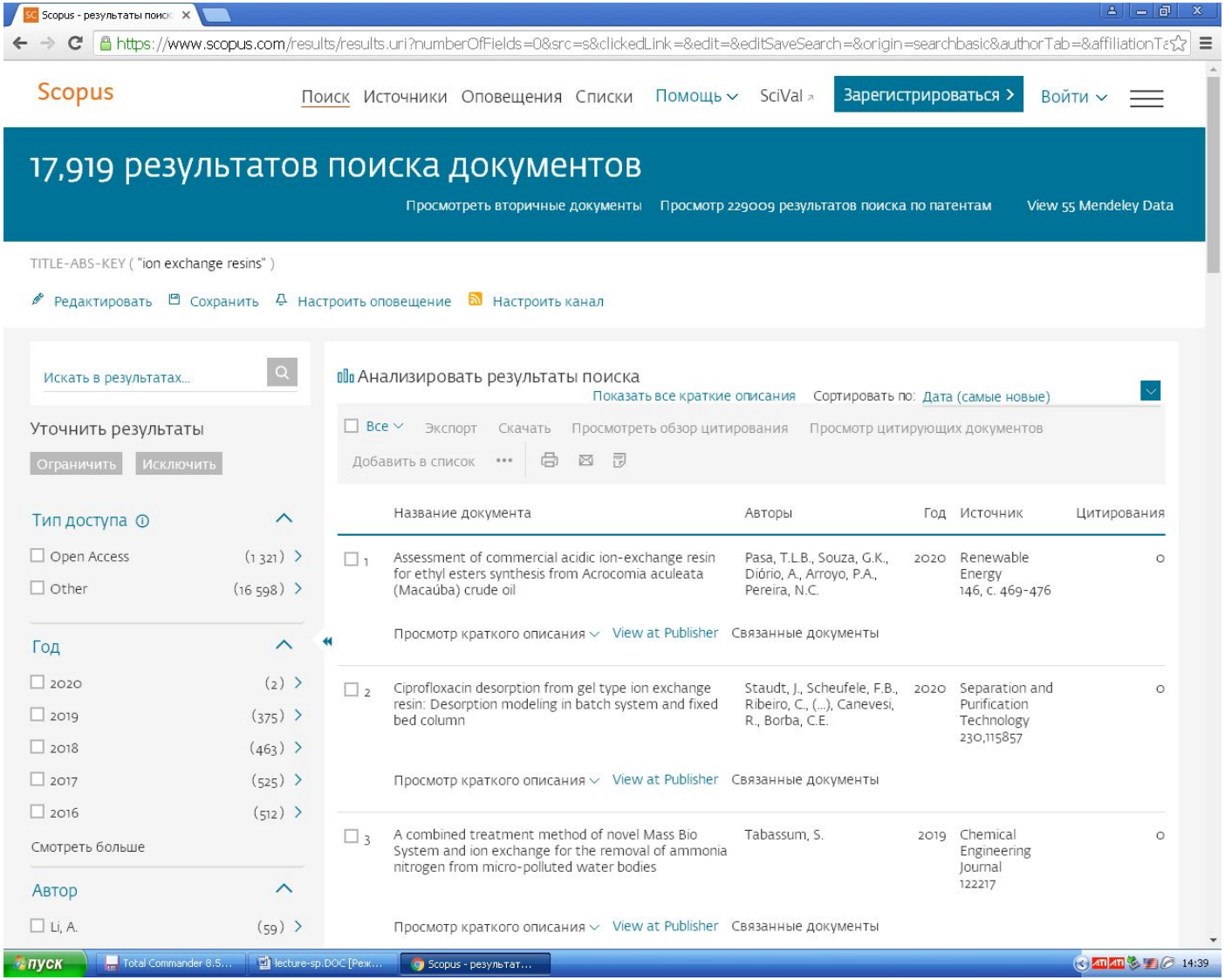

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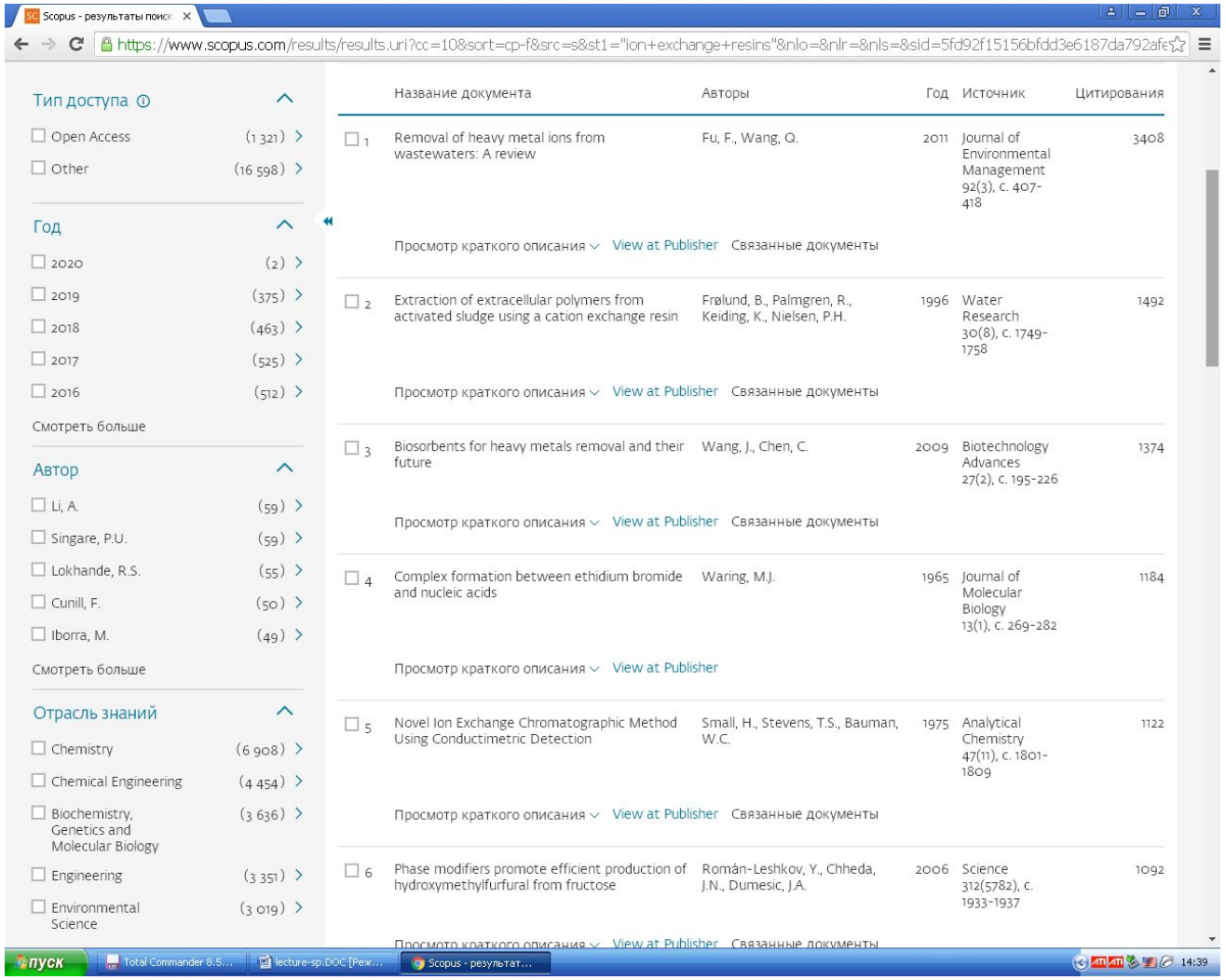

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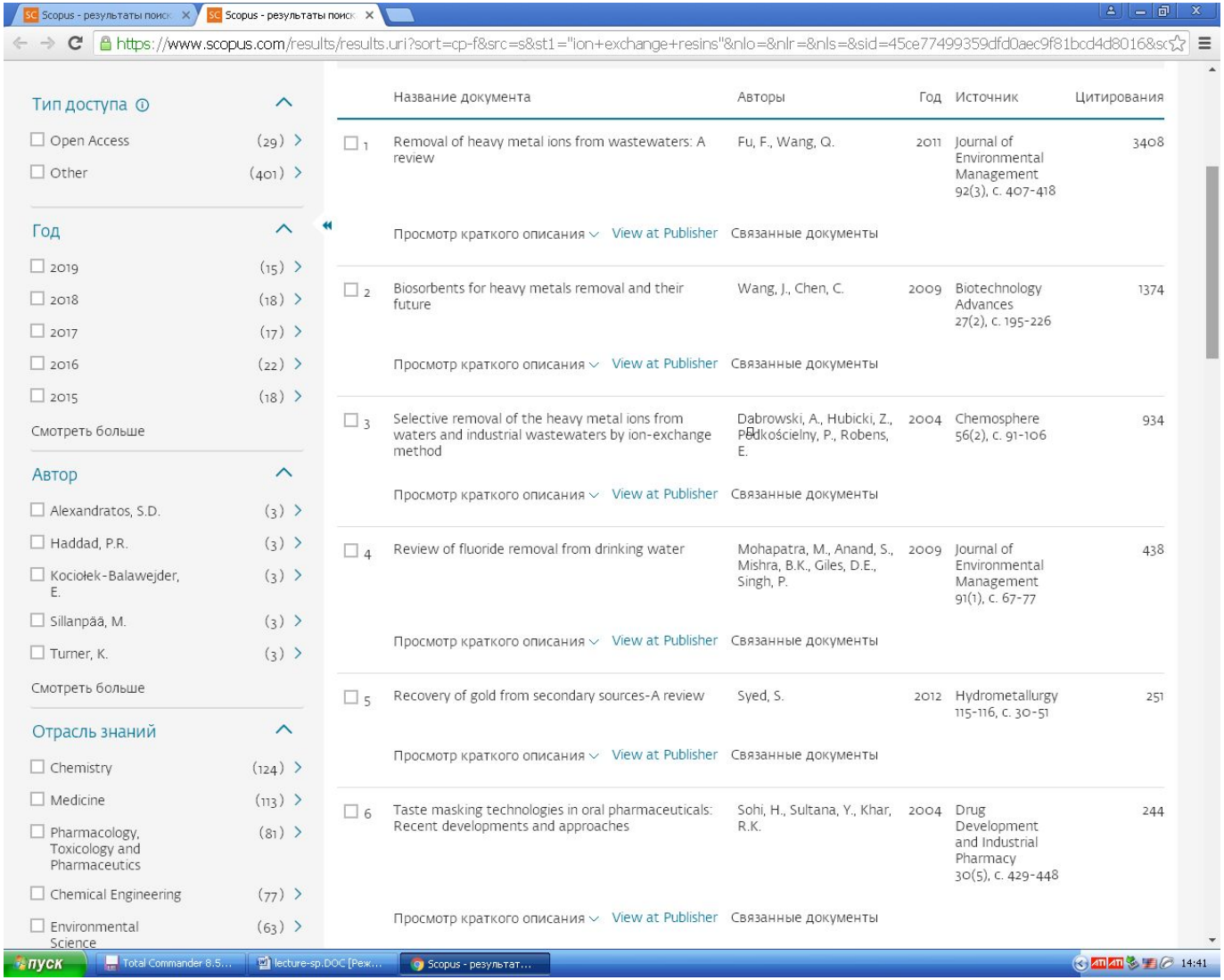

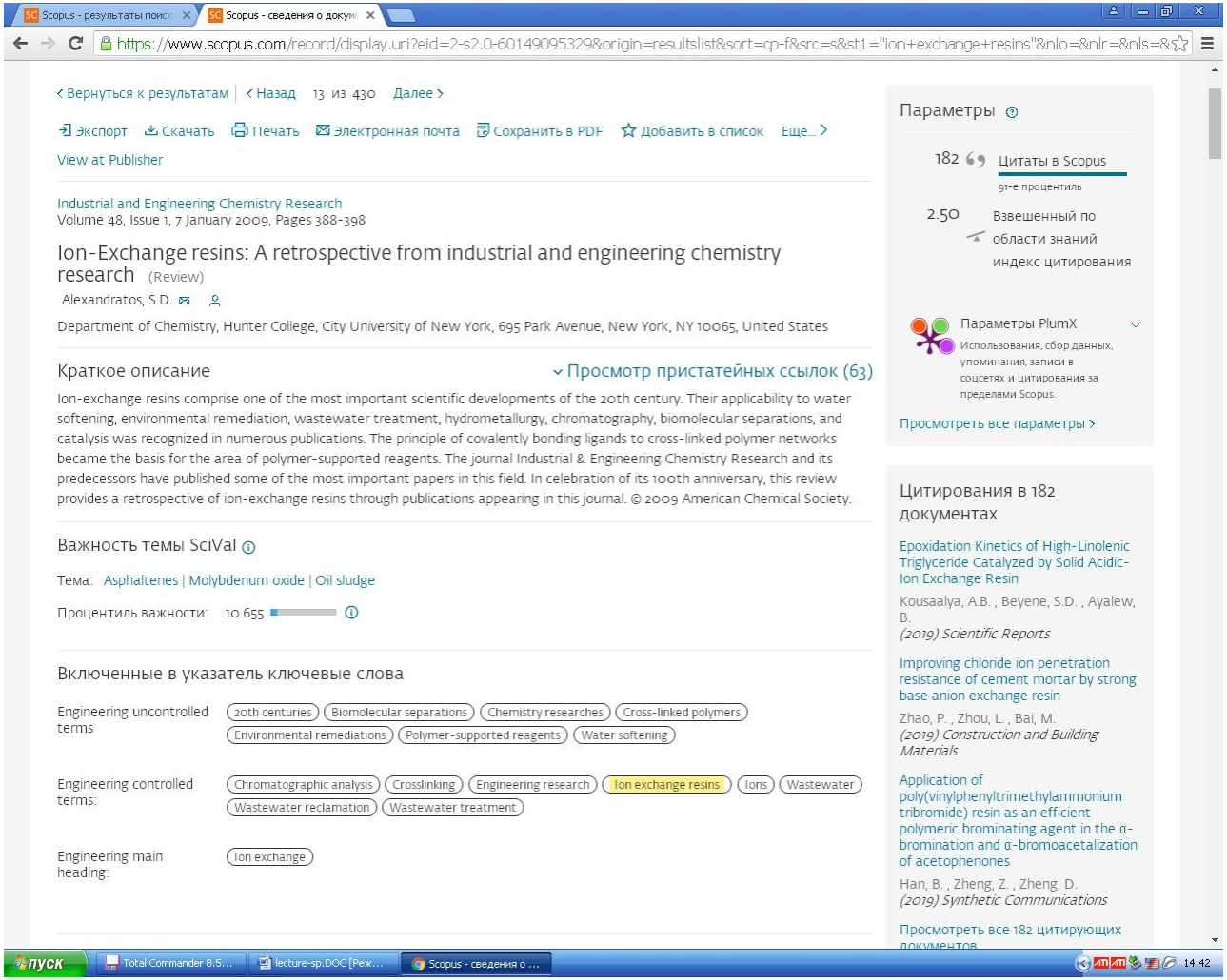
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Ion-Exchange resins: A retrospective from industrial and engineering chemistry research (Review)
Alexandratos, S.D.
Department of Chemistry, Hunter College, City University of New York, 695 Park Avenue, New York, NY 10065, United States
Краткое описание
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Ion-exchange resins comprise one of the most important scientific developments of the 20th century. Their applicability to water softening, environmental remediation, wastewater treatment, hydrometallurgy, chromatography, biomolecular separations, and catalysis was recognized in numerous publications. The principle of covalently bonding ligands to cross-linked polymer networks became the basis for the area of polymer-supported reagents. The journal Industrial & Engineering Chemistry Research and its predecessors have published some of the most important papers in this field. In celebration of its 100th anniversary, this review provides a retrospective of ion-exchange resins through publications appearing in this journal. © 2009 American Chemical Society.
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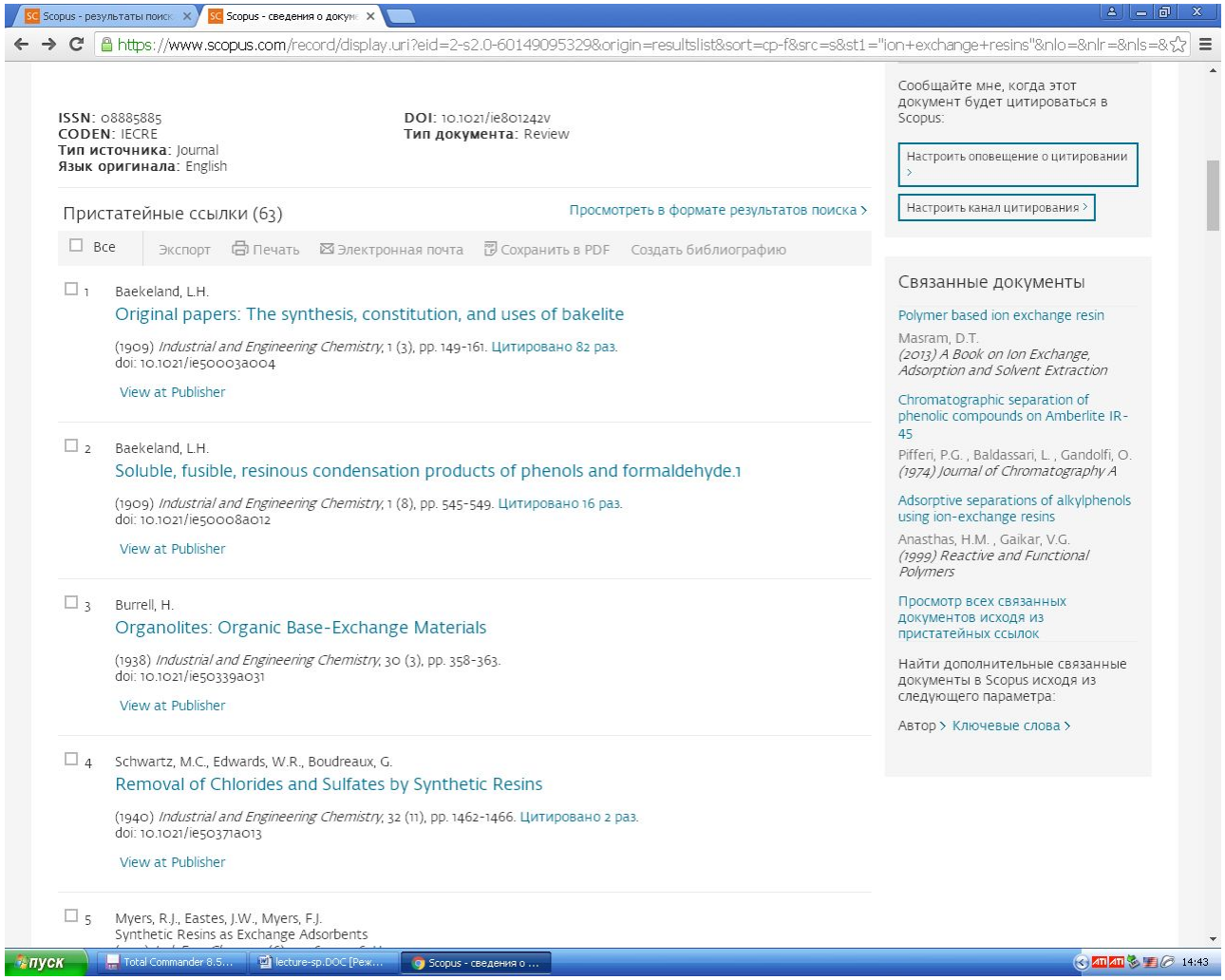

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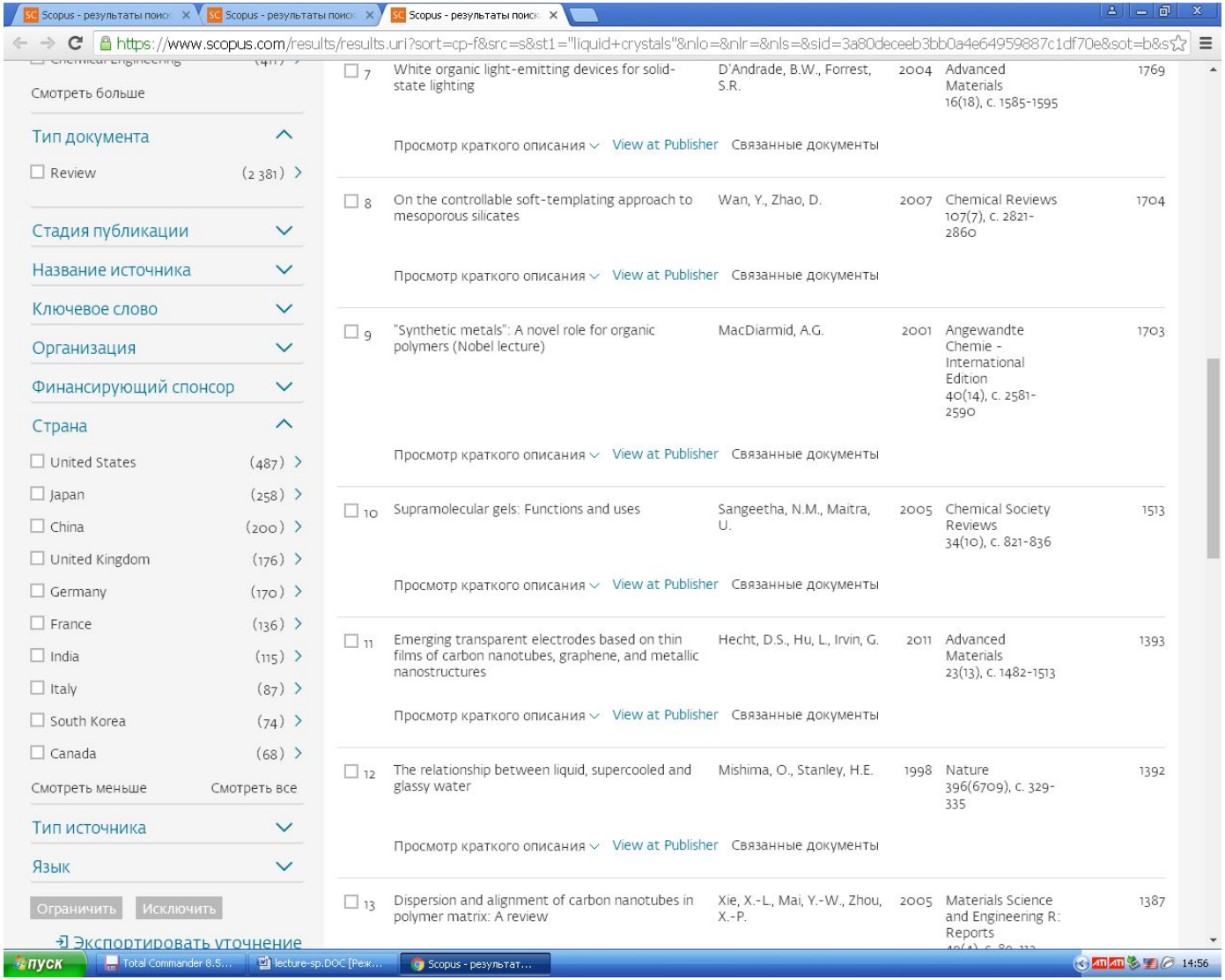

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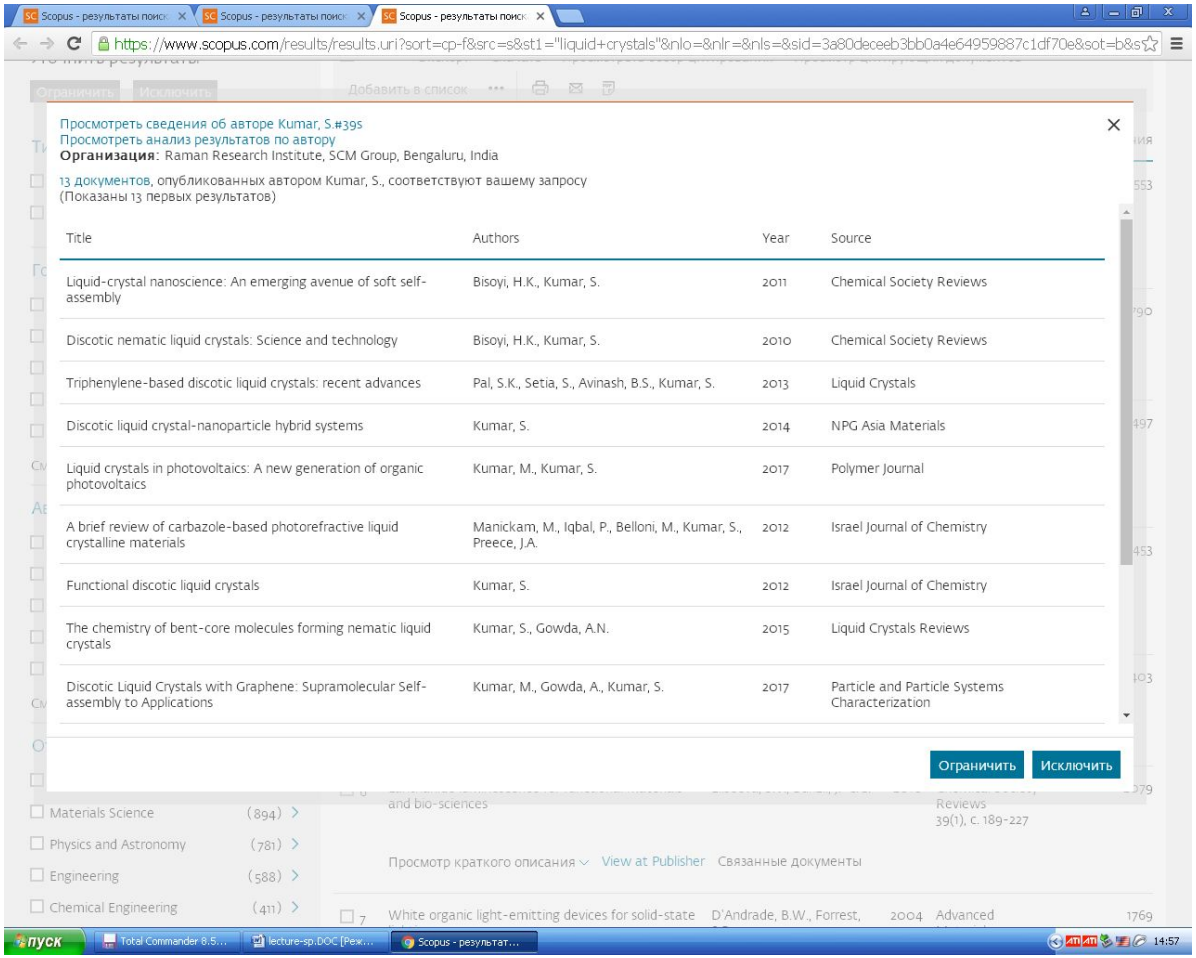

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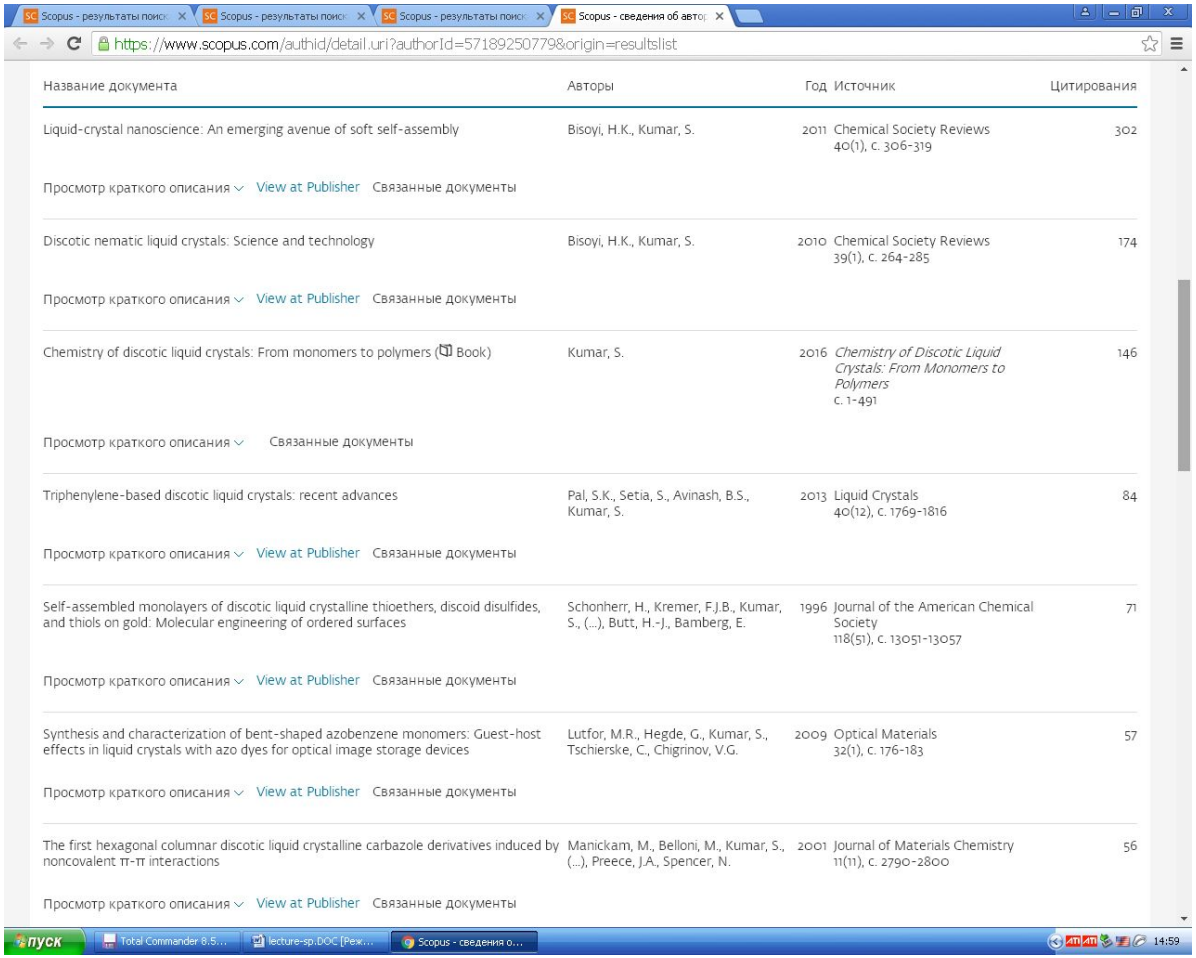

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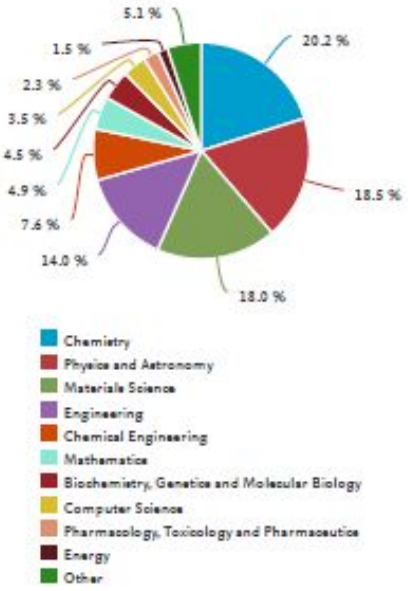

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