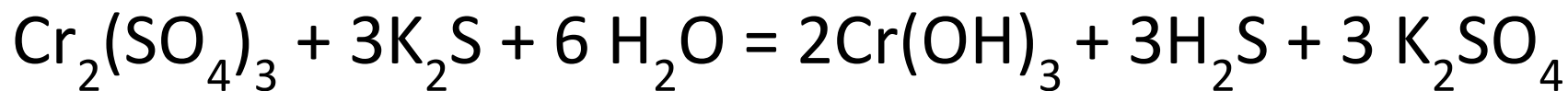
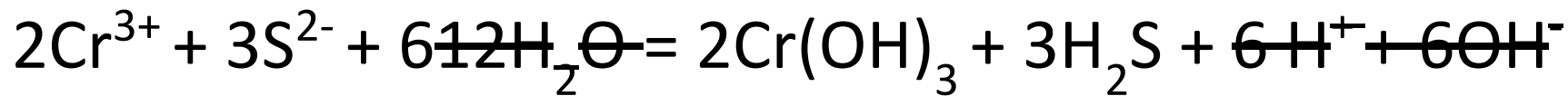
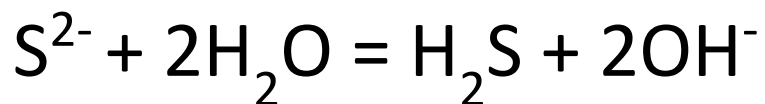
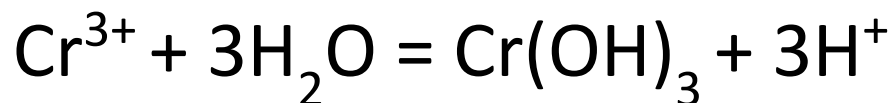
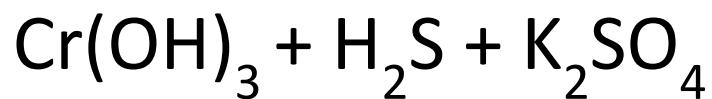
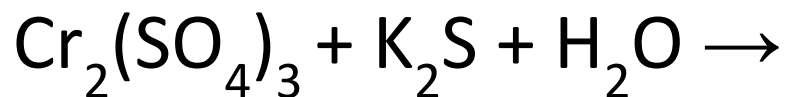


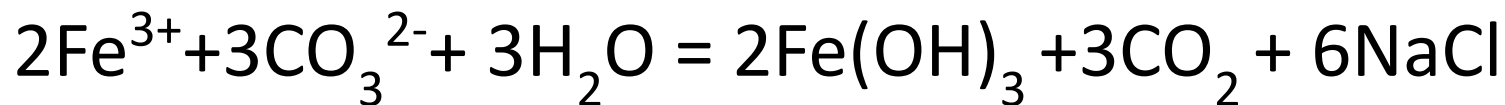
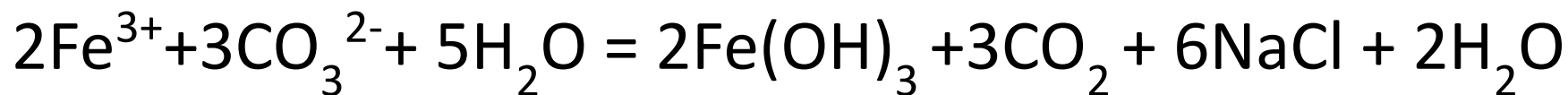
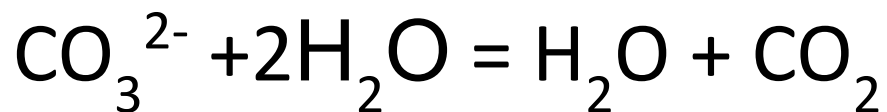
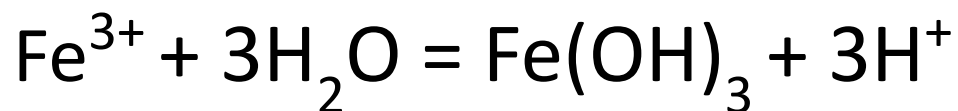
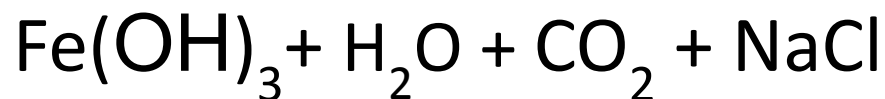
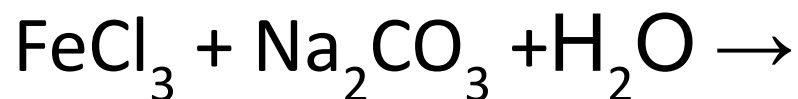
**Необратимый
(совместный)
гидролиз**

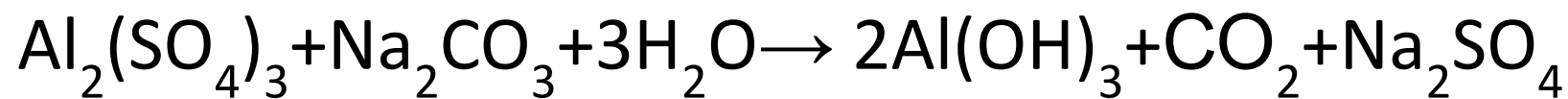
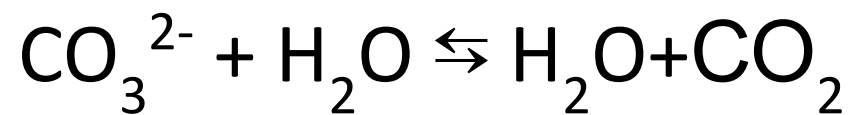
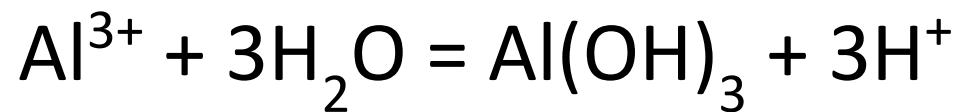
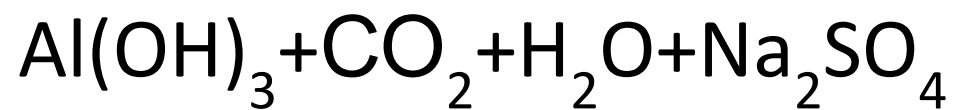
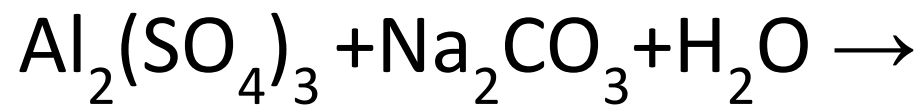
Соль слабого основания сильной кислоты
+ соль сильного основания слабой
кислоты + H_2O

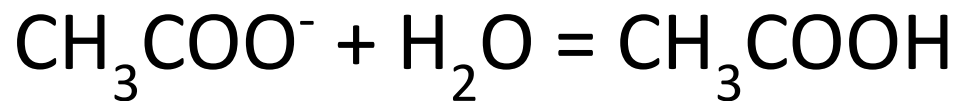
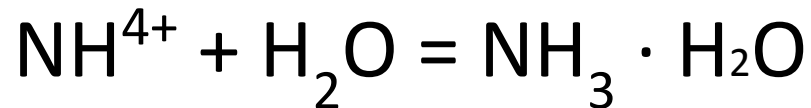
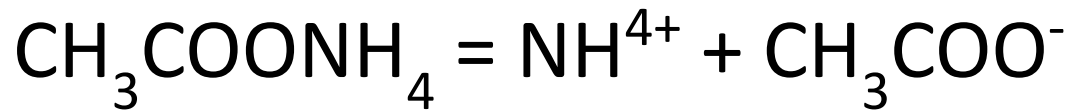
→

слабое основание + слабая кислота + соль
сильного основания и сильной кислоты









- $\text{Cr}_2\text{S}_3 + 6\text{H}_2\text{O} = 2\text{Cr}(\text{OH})_3\downarrow + 3\text{H}_2\text{S}$
- $(\text{NH}_4)_3\text{PO}_4 + \text{H}_2\text{O} =$
- $(\text{NH}_4)_2\text{S} + \text{H}_2\text{O} =$
-