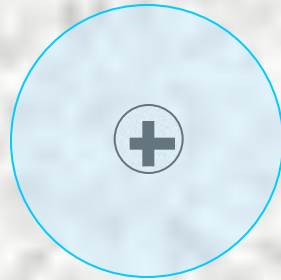
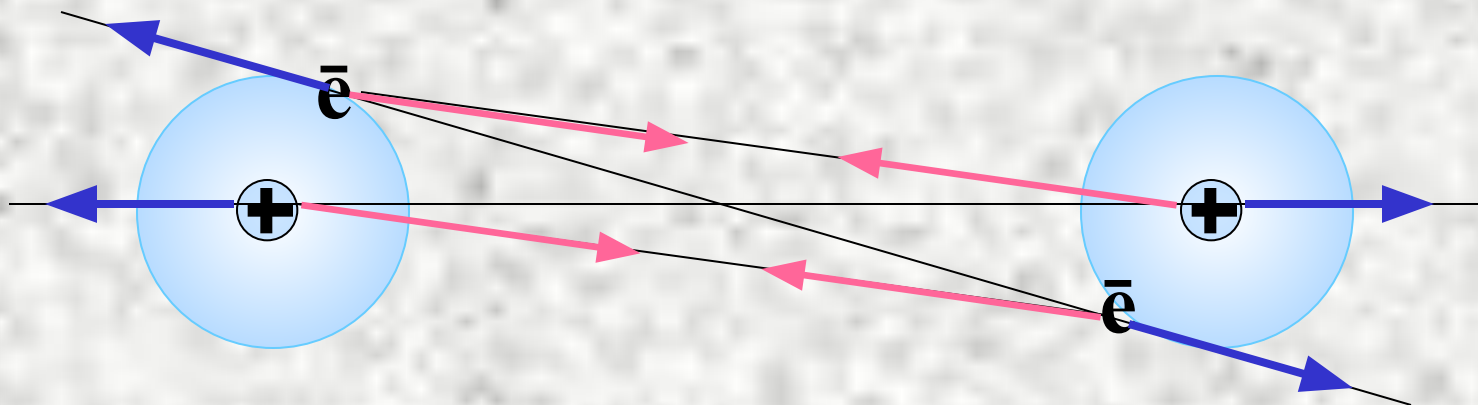


ХИМИЧЕСКАЯ СВЯЗЬ

- совокупность сил, обуславливающих взаимодействие атомов друг с другом в химическом соединении.

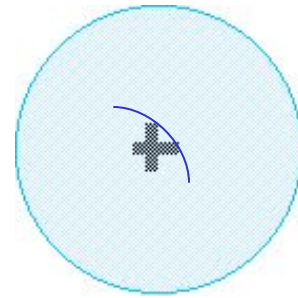
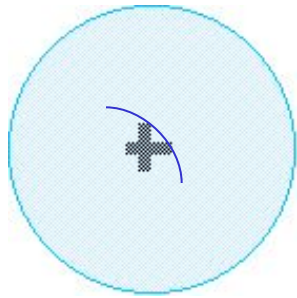




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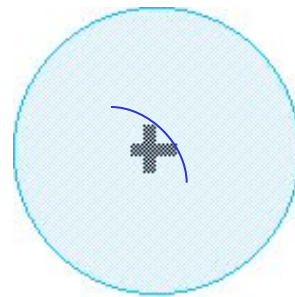
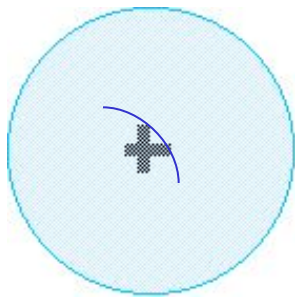


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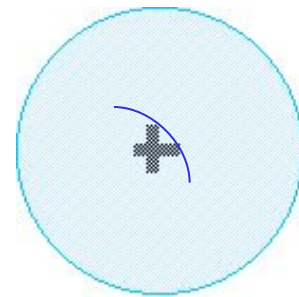
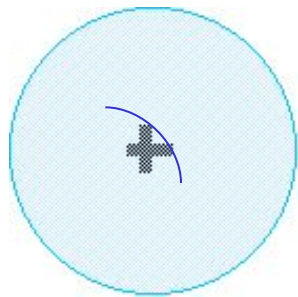


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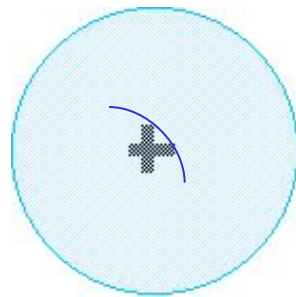
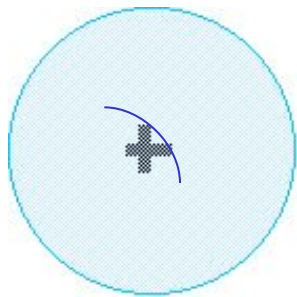


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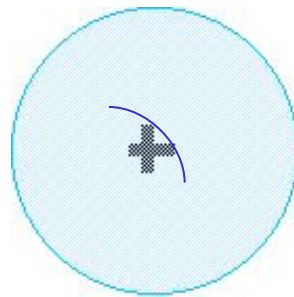
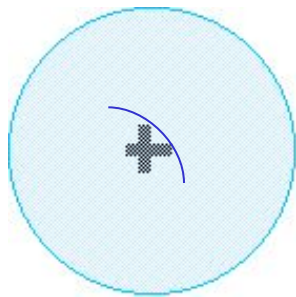


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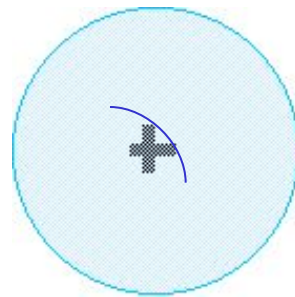
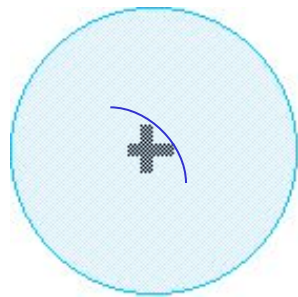


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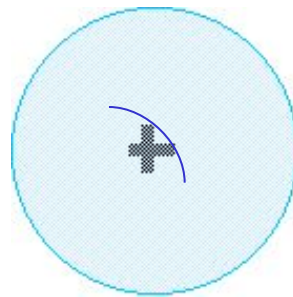
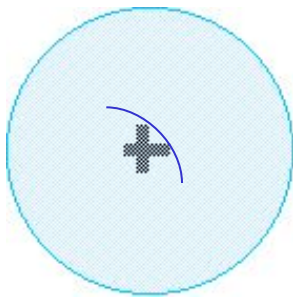


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F_{отт.}



F_{пр.}



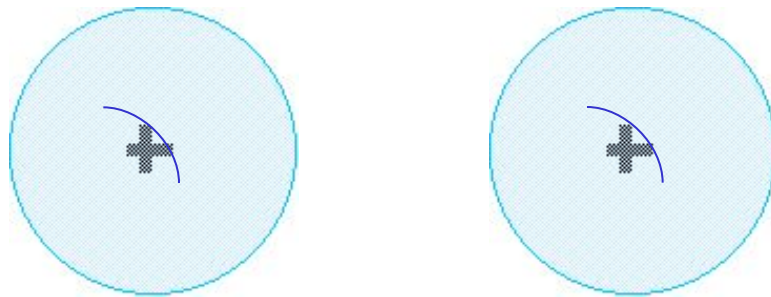


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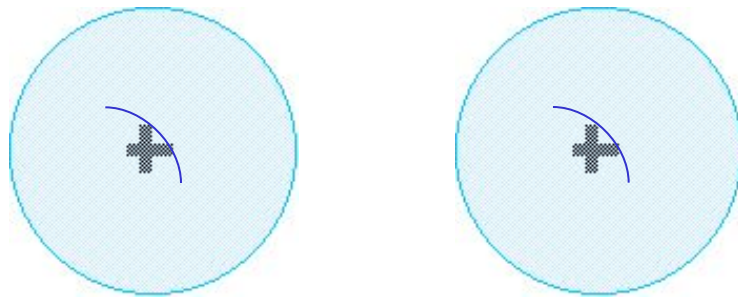


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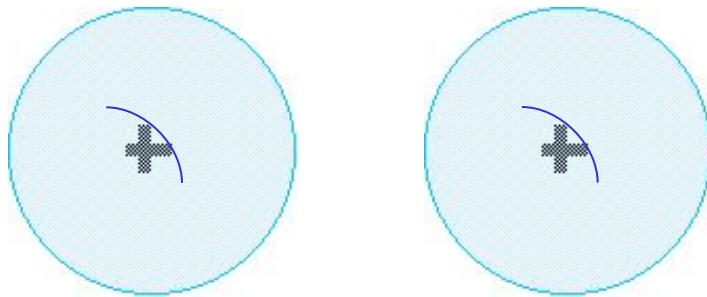


F_{отт.}



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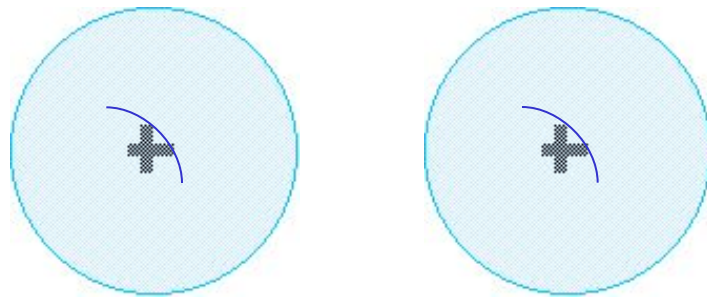


F_{отт.}



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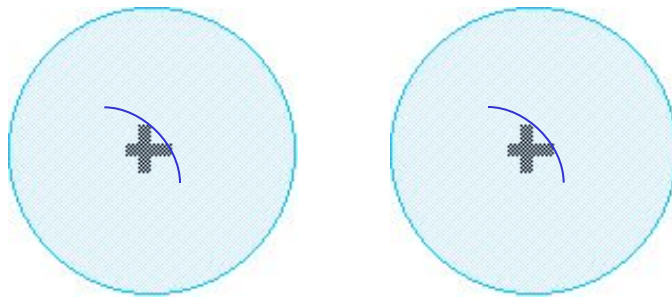


F_{отт.}



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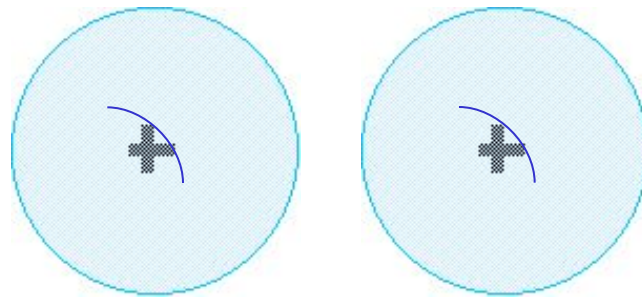


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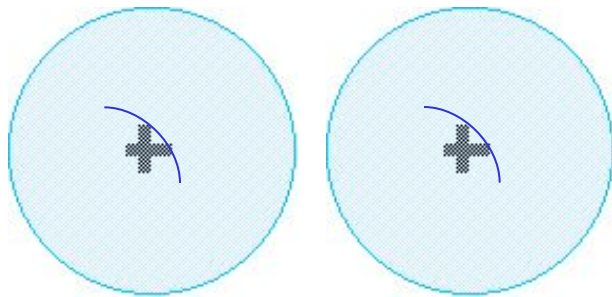


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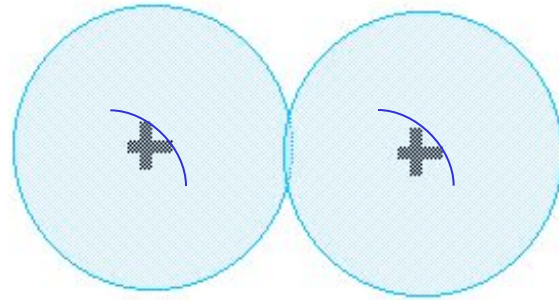


F_{отт.}



F_{пр.}



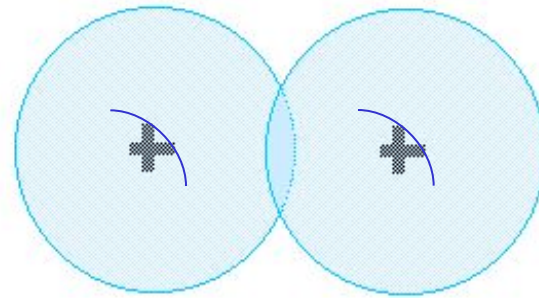


F_{отт.}



F_{пр.}



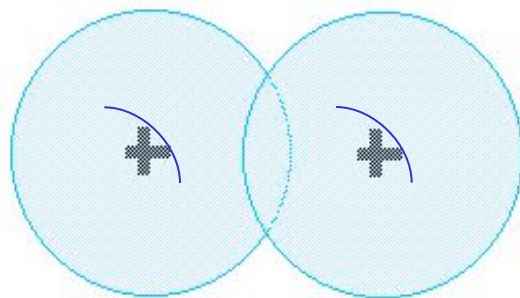


F_{отт.}



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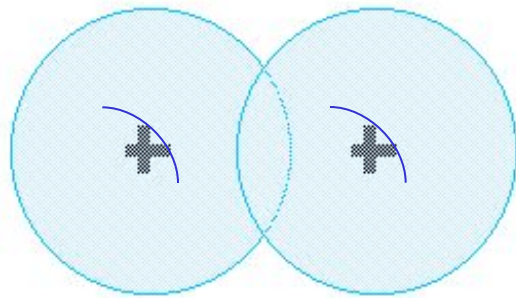


F_{отт.}



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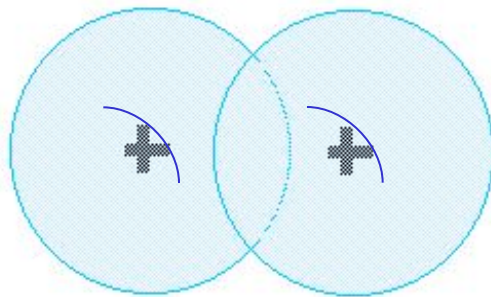


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F_{пр.}





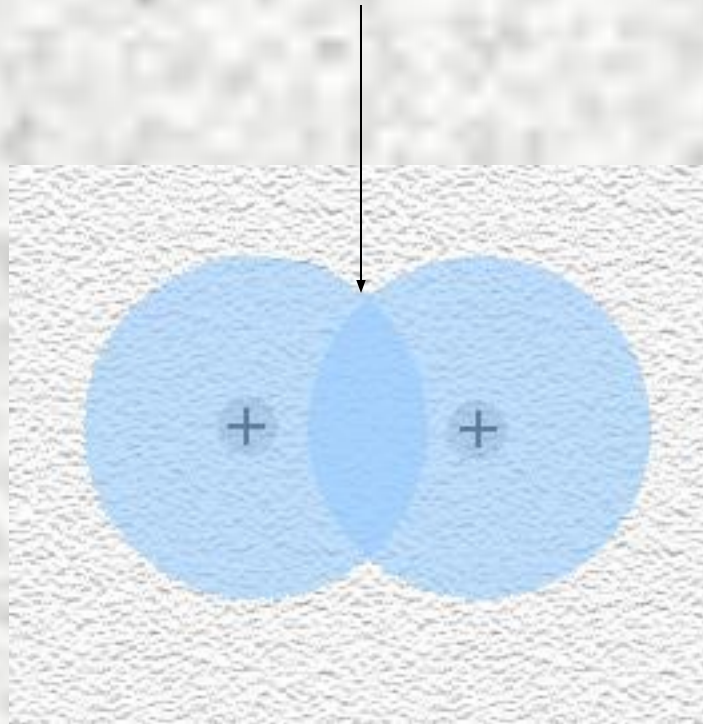
F_{отт.}

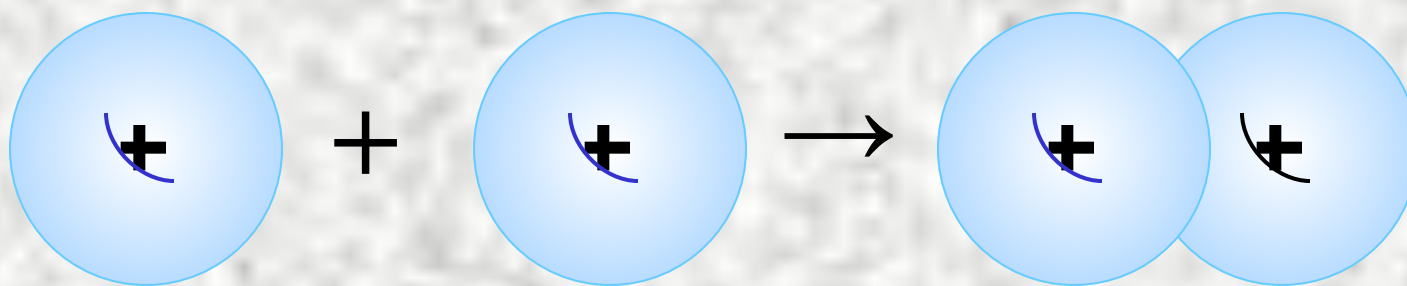


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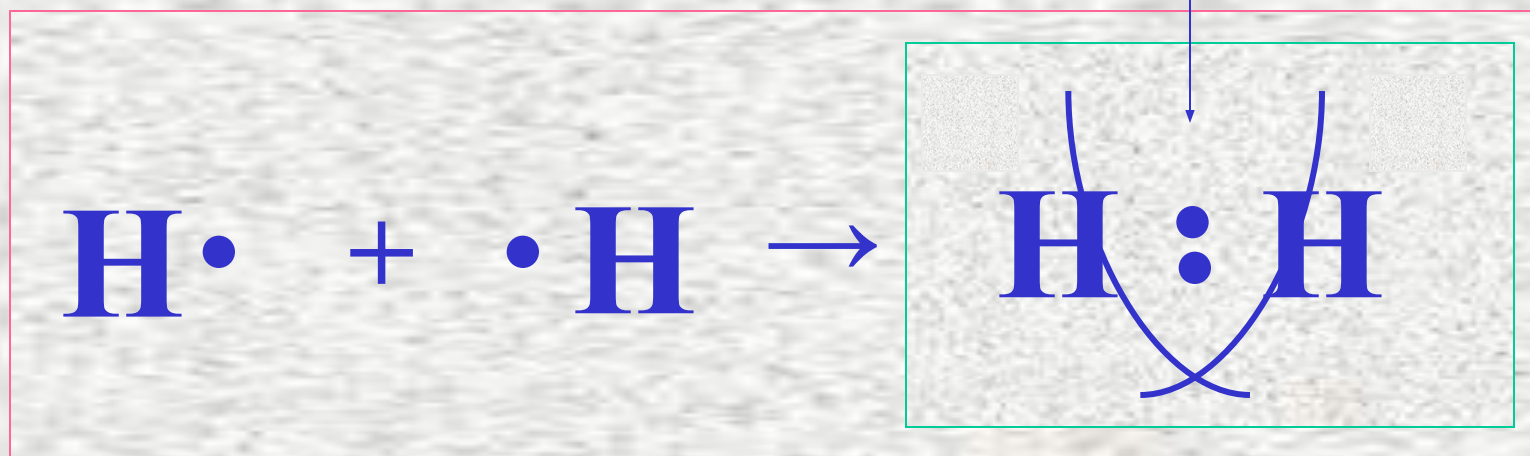


Область нахождения двух электронов —
наибольшая электронная плотность



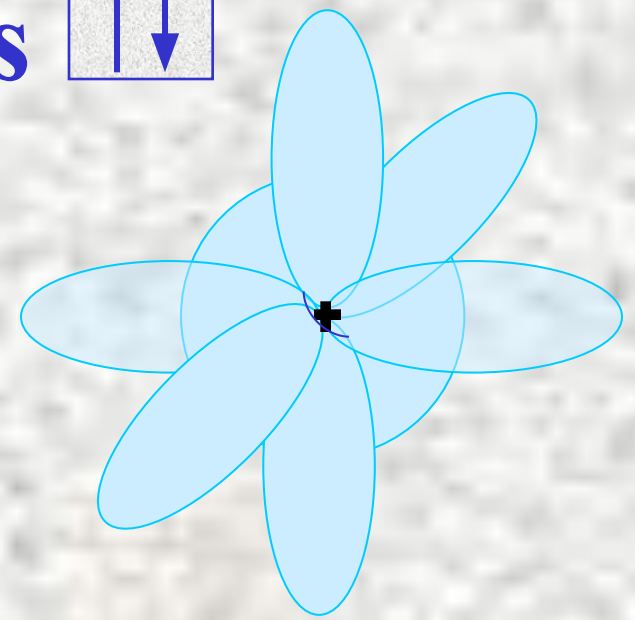
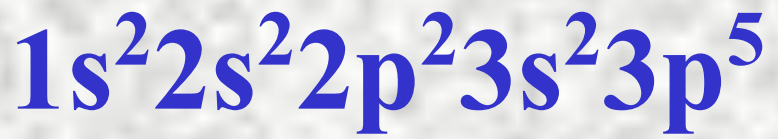
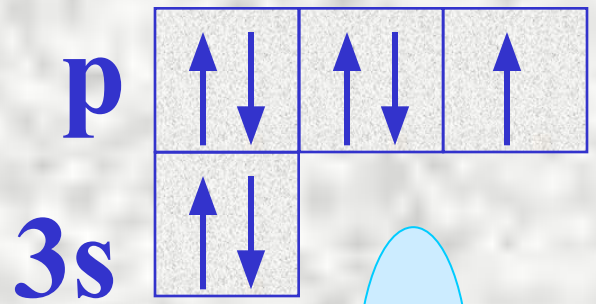
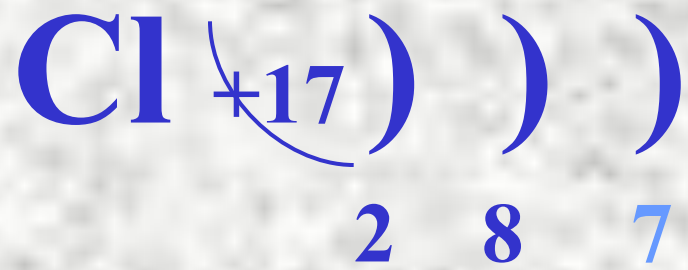
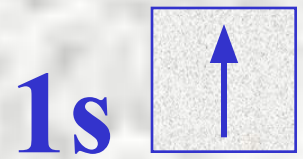


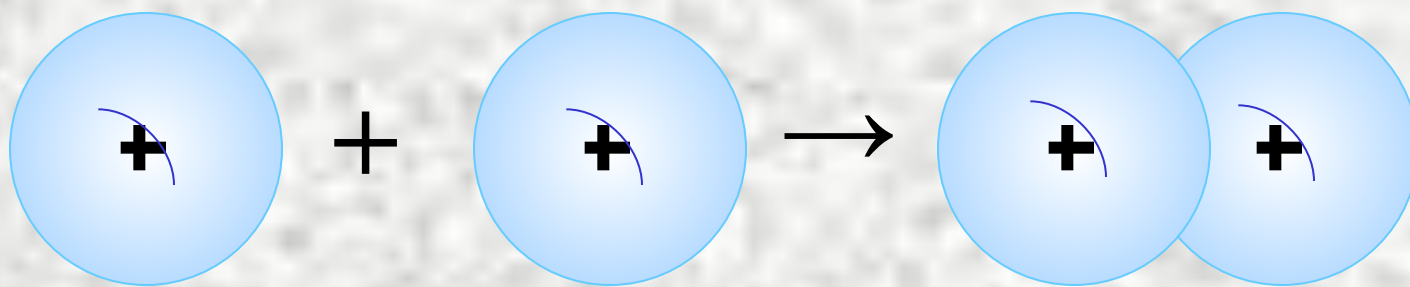
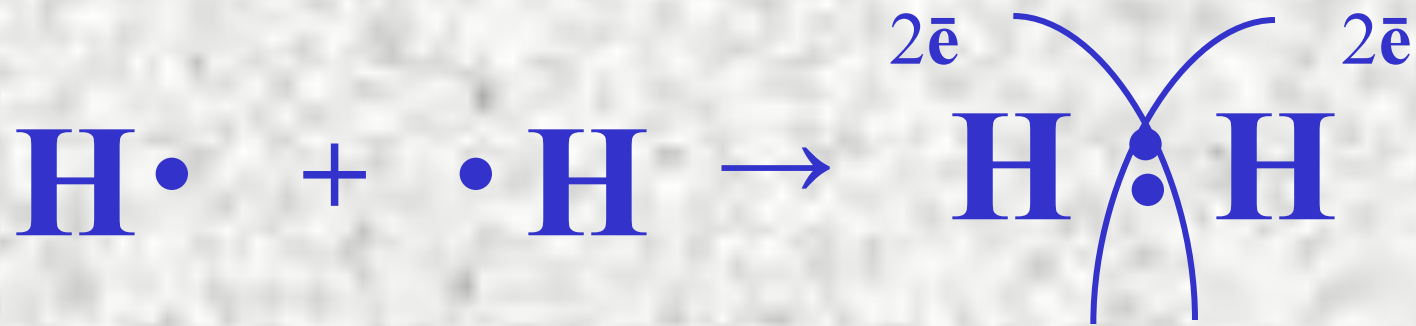
*одна общая электронная пара –
одна химическая связь*

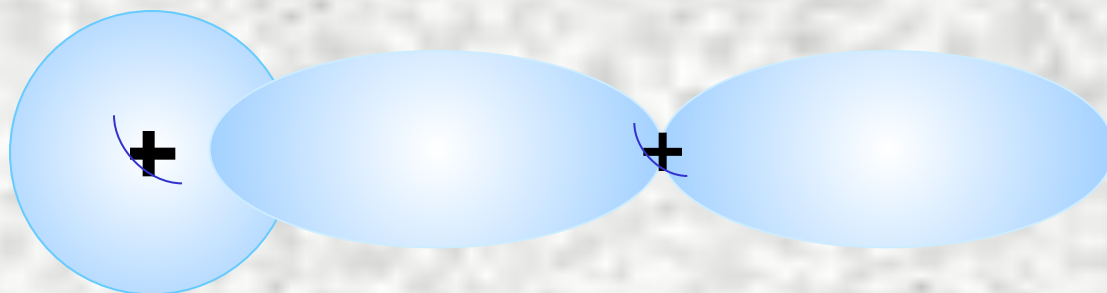
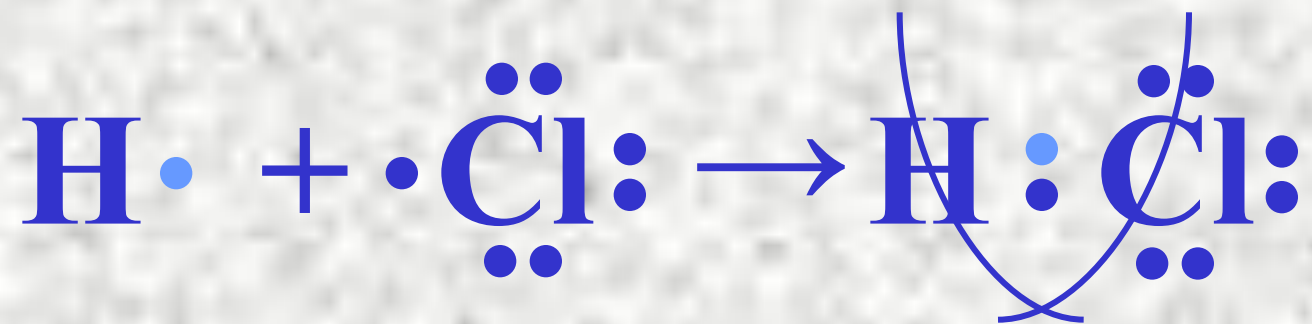


*электронная схема
образования молекулы*

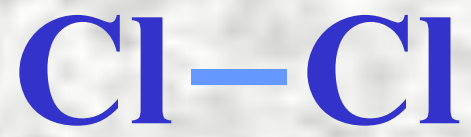
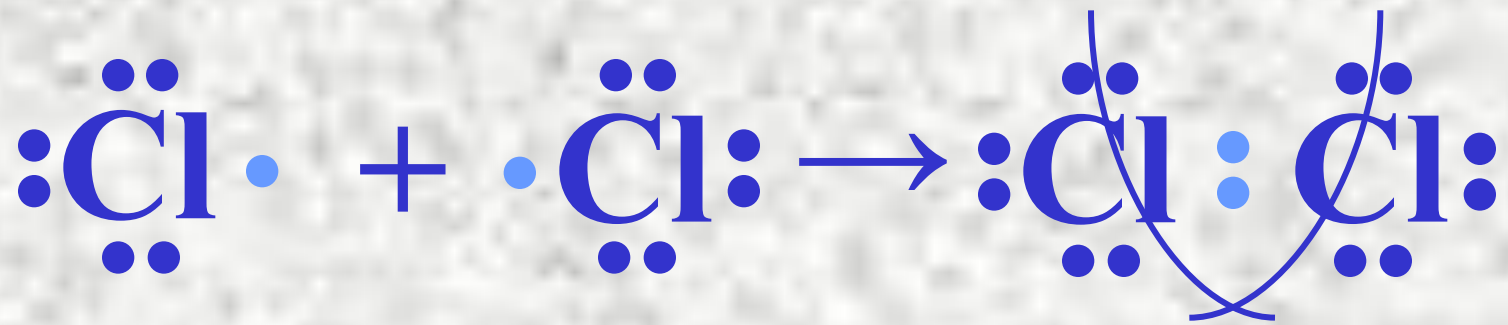
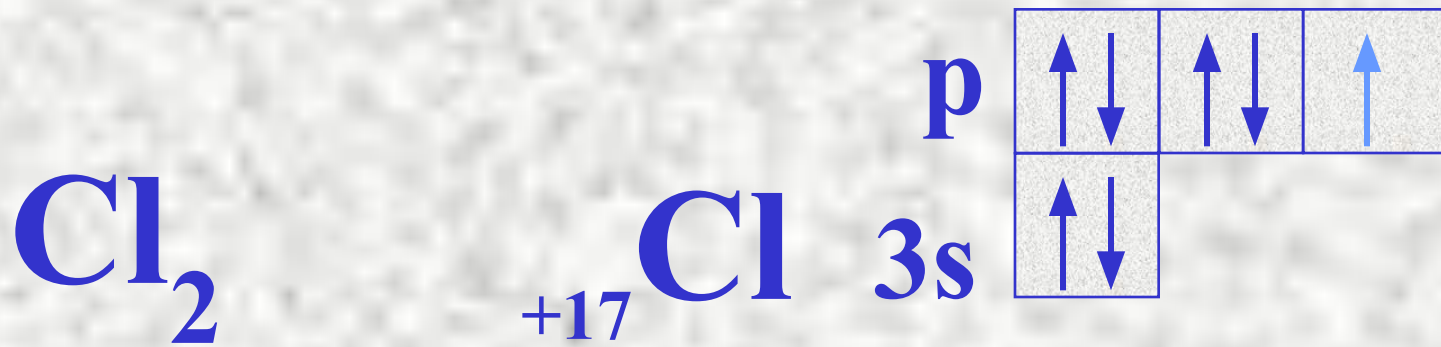
*электронная формула
молекулы*



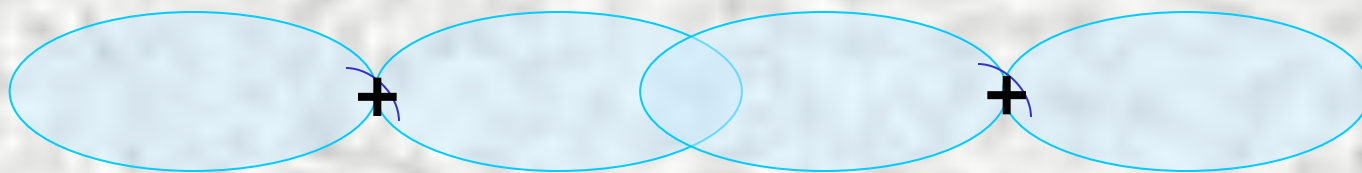




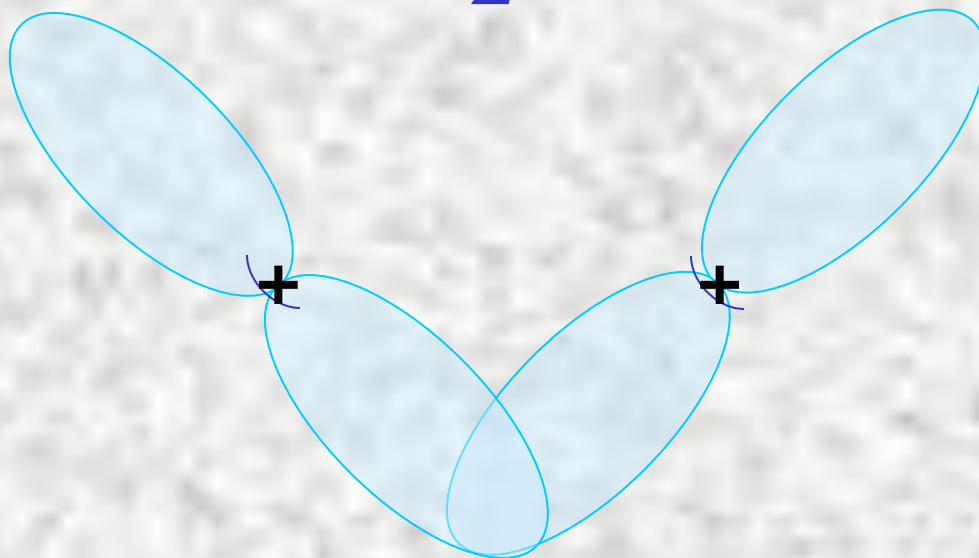
Ковалентная связь — это связь между атомами, возникающая за счёт образования общих электронных пар.



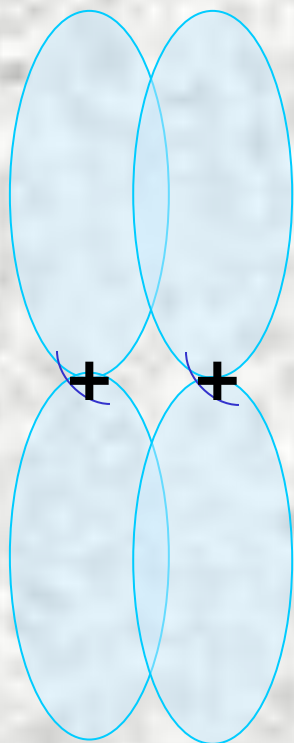
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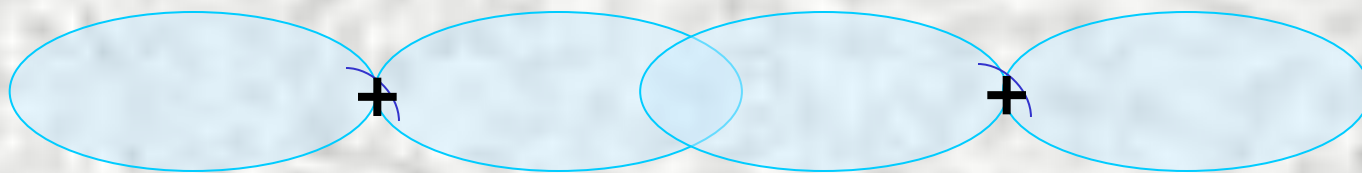
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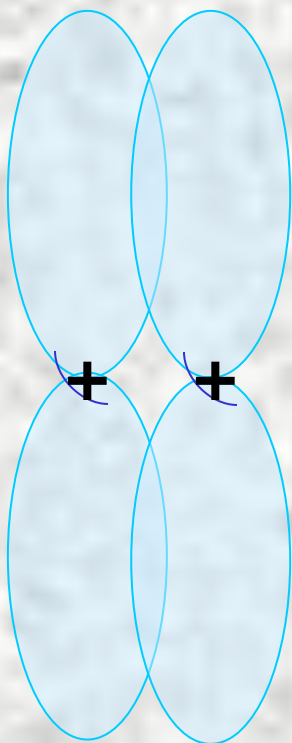


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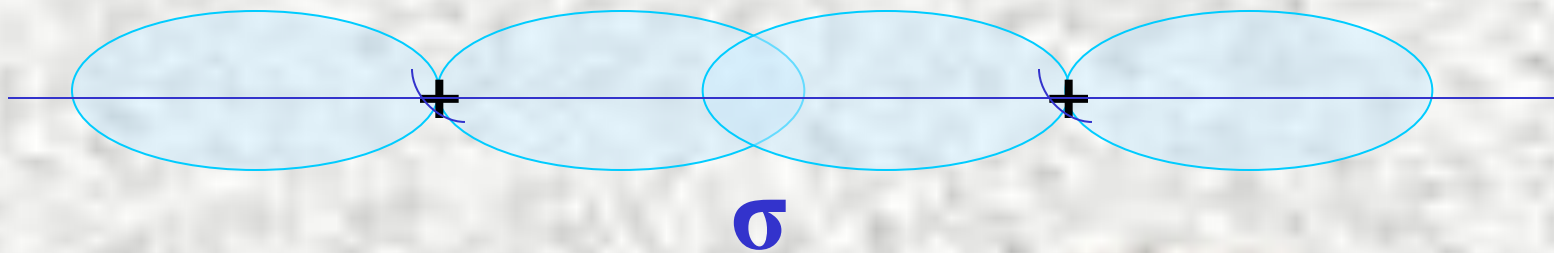
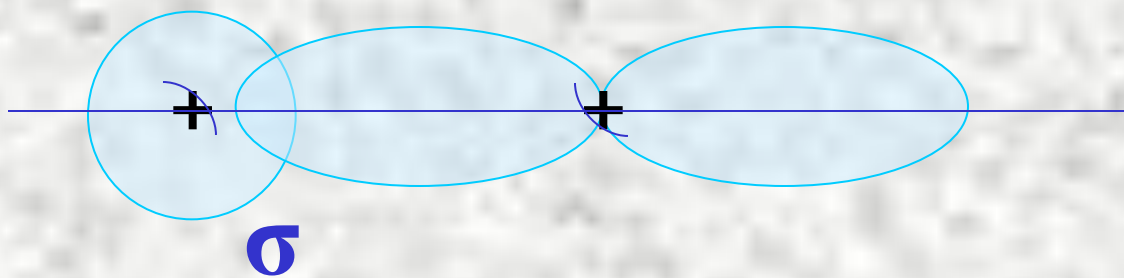
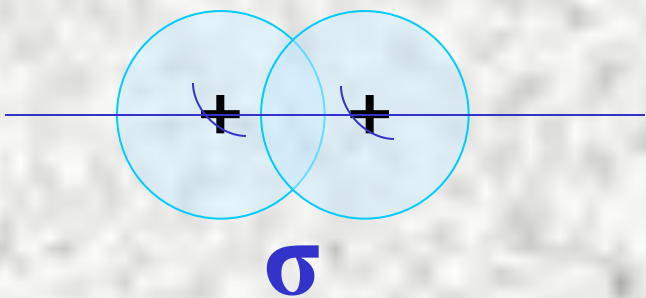
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σ-связь — ковалентная связь, при образовании которой область перекрывания атомных орбиталей находится на линии, соединяющей центры взаимодействующих атомов.

Между парой атомов может быть только **одна σ-связь**.