

# Study of Short Peptide Adsorption on Solution Dispersed Inorganic Nanoparticles Using Depletion Method

Elena Korina<sup>a</sup>, Sergei Naifert<sup>a</sup>, Roman Morozov<sup>a</sup>,  
Vladimir Potemkin<sup>a,b</sup>, Oleg Bol'shakov<sup>a\*</sup>

<sup>a</sup> Nanotechnology Education and Research Center, South Ural State University, 454080 Chelyabinsk, Russia

<sup>b</sup> Laboratory of Computational Modelling of Drugs, South Ural State University, 454080 Chelyabinsk, Russia

## Preparation of 16 mM Dipeptide Stock Solution

---

# Preparation of Peptide Dilutions

---

# Preparation of Titania Sol

---

# Mixing and Thermostating



# Filtration of the Thermostated Samples

---

# Preparation of Derivatization Solutions

---

# Derivatization

---



# HPLC Analysis

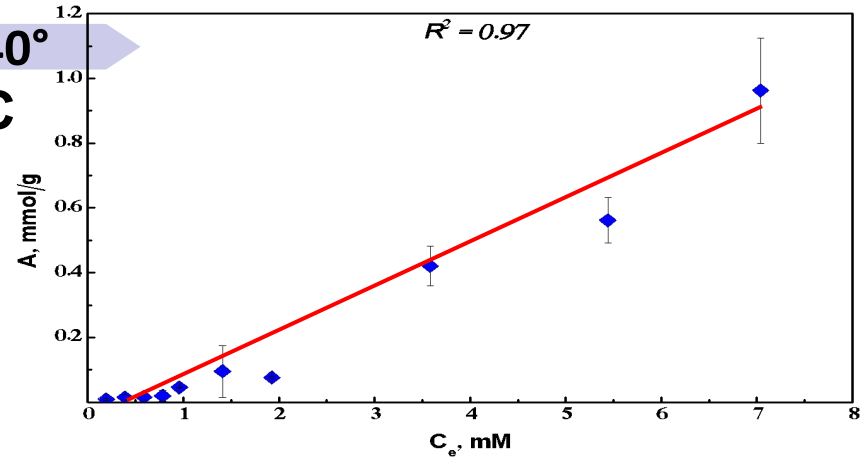
---

## Representative results

---

40°

C

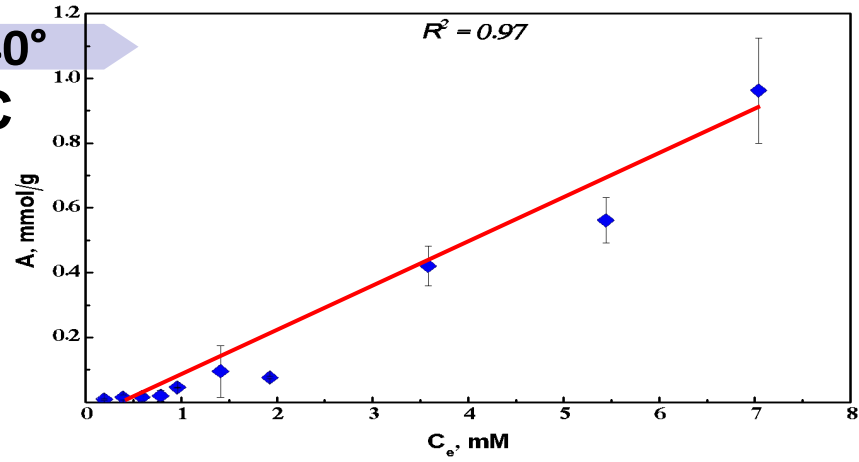


$$A = K_H C_e$$

$$K_H = \frac{dA}{dC_e}$$

40°

C



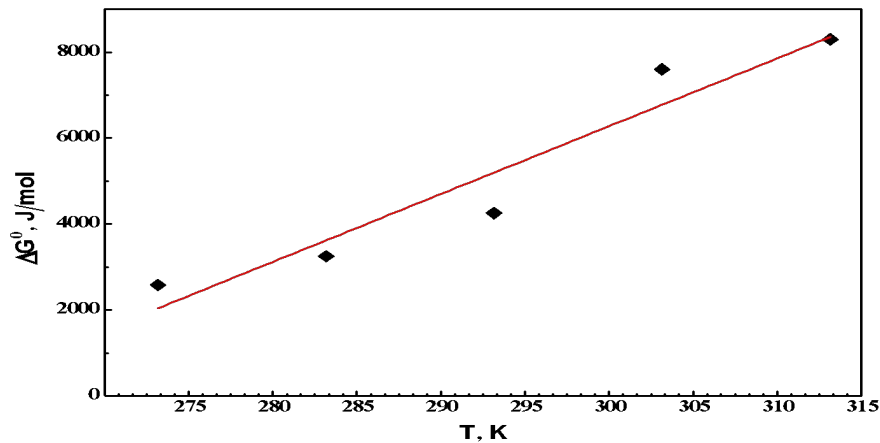
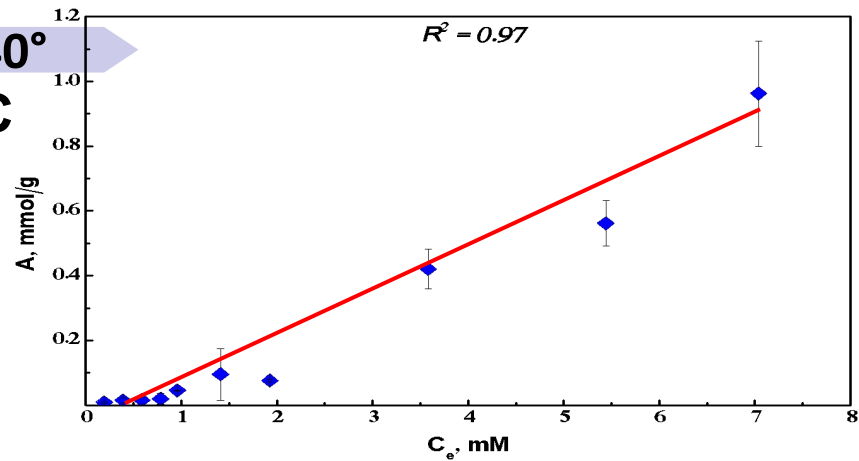
$$A = K_H C_e$$

$$K_H = \frac{dA}{dC_e}$$

$$\Delta G = -RT \ln K$$

40°

C



$$A = K_H C_e$$

$$K_H = \frac{dA}{dC_e}$$

$$\Delta G = -RT \ln K$$

$$\Delta G = \Delta H - T\Delta S$$

$$\Delta G = -41000 + 160 T$$

# Physico-chemical Constants of Ile-His Adsorption on $\text{TiO}_2$

Table 1

T, K	$K_H$	$\Delta G^0$ , kJ/mol	$\Delta H^0$ , kJ/mol	$\Delta S^0$ , kJ/mol K
273.15	0.32	2.6	-41	
283.15	0.25	3.2		
293.15	0.17	4.3		-0.16
303.15	0.05	7.6		
313.15	0.04	8.3		

# Conclusions

---

**Executive producer**

Sergei Gordienko

**Producer**

Dmitry Shytyakov

**Scriptwriter**

Oleg Bol'shakov

**Videographer**

Pavel Shytyakov

**Editors**

Roman Morozov

Elena Korina

---