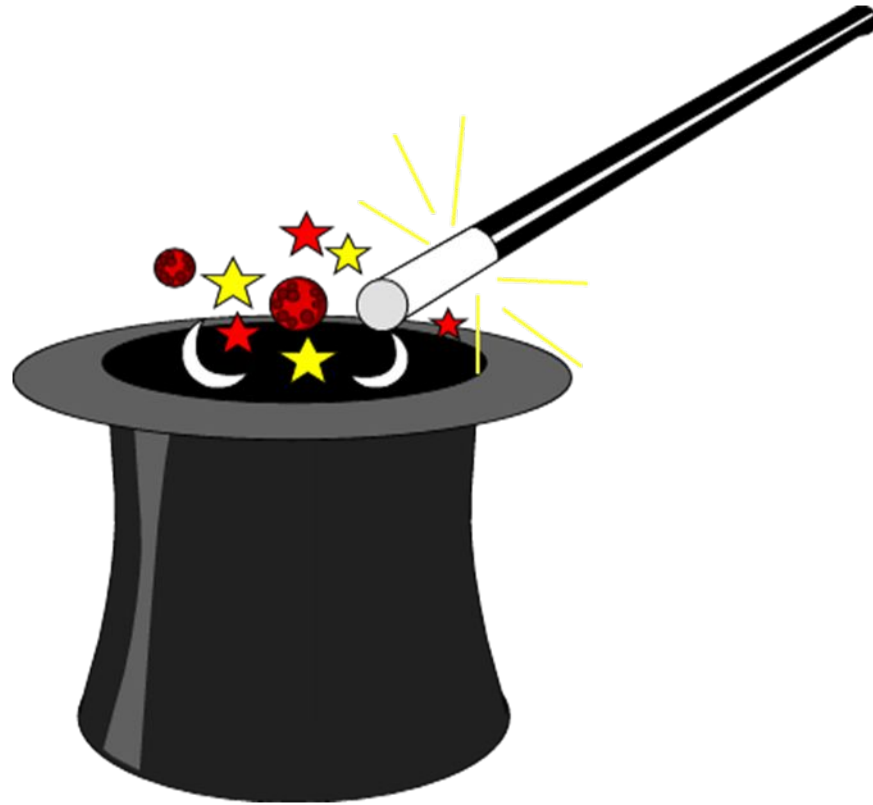




- **ABG CASE  
STUDIES &  
INTERPRETATION**

It's not magic understanding ABG's,  
it just takes a little practice!



# Acid-base imbalances

- Metabolic acidosis
- Metabolic alkalosis
- Respiratory acidosis
- Respiratory alkalosis

# Metabolic

- **METABOLIC ACIDOSIS:** Decrease the  $\text{HCO}_3^-$  --> the pH goes down.
- **Compensation:** Respiratory Alkalosis (hyperventilation) will bring the pH back near normal.
- Causes: Diarrhea, DKA, LA, renal failure.
- **METABOLIC ALKALOSIS:** Increase the  $\text{HCO}_3^-$  --> the pH goes up.
- **Compensation:** Respiratory Acidosis (hypoventilation) can help to bring the pH+

# Respiratory

- **RESPIRATORY ACIDOSIS**: Increase the  $\text{PCO}_2$ ---> the pH goes down.  
*Hypoventilation*. **Compensation: Metabolic Alkalosis** can help bring the pH back near normal.
- **Causes**: pneumonia, Bronchitis, Asthma
- **RESPIRATORY ALKALOSIS**: Decrease the  $\text{PCO}_2$ -> the pH goes up.  
*Hyperventilation*.
- **Compensation: Metabolic Acidosis** can
- help bring the pH back near normal.

# METABOLIC ALKALOSIS

## CAUSES:

- Vomiting: Lose enough stomach acid to produce alkalosis.
- Diuretics: Loop diuretics and thiazides can lead to hypokalemia -----> secondary metabolic alkalosis.
- Antacids overuse

# RESPIRATORY ACIDOSIS:

## causes:

### **CNS DEPRESSION**

- DRUGS: Opiates, sedatives, anaesthetics
- OBESITY HYPOVENTILATION SYNDROME
- STROKE

### **NEUROMUSCULAR DISORDERS:**

- NEUROLOGIC: POLIO, GBS, TETANUS, BOTULISM
- MUSCULAR DYSTROPHY

### **AIRWAY OBSTRUCTION**

- ACUTE ASPIRATION, LARYNGOSPASM

### **CHEST WALL RESTRICTION**

- PLEURAL: Effusions, empyema, pneumothorax, fibrothorax
- CHEST WALL: Kyphoscoliosis, scleroderma, ankylosing spondylitis, obesity

### **SEVERE PULMONARY RESTRICTIVE DISORDERS**

PULMONARY FIBROSIS

# RESPIRATORY ALKALOSIS

- **Causes:**
- High altitude.
- Neuromuscular disease
- Respiratory center depression
- Inadequate mechanical ventilation
- Sepsis
- Burns



# Metabolic acidosis

- **Metabolic acidosis:** Is caused by a decrease in  $\text{HCO}_3^-$  concentration in blood.
- Causes:
  1. **Increased production of acids:** LA, kA, Salicylate poisoning.
  2. **Loss of  $\text{HCO}_3^-$ :** Diarrhea and kidneys RTA.
  3. **Blood profile:** pH decreased
    - $[\text{HCO}_3^-]$  decreased,  $\text{PCO}_2$  decreased

# Compensation of Metabolic acidosis:

- Respiratory compensation: decrease in pH stimulates respiratory center causing hyperventilation which produces decrease in  $PCO_2$ .
- Renal Compensation: excess  $H^+$  is excreted as titratable acid and  $NH_4^+$ .
- Treatment: lactate containing solution which converts  $HCO_3^-$  ion the liver.

# ABG Disorders

Disorder	Change	Compensation
Respiratory Acidosis	↑ Pa CO <sub>2</sub>	↑ HCO <sub>3</sub> (Metabolic alkalosis)
Respiratory Alkalosis	↓ Pa CO <sub>2</sub>	↓ HCO <sub>3</sub> (Metabolic acidosis)
Metabolic Acidosis	↓ HCO <sub>3</sub>	↓ Pa CO <sub>2</sub> ↑ (Respiratory alkalosis)
Metabolic Alkalosis	↑ HCO <sub>3</sub>	↑ Pa CO <sub>2</sub> (Respiratory acidosis)

# Assessment of acid base status

- Direct arterial blood measurements: ABG

pH  
pCO<sub>2</sub>  
pO<sub>2</sub>

NB: use heparinised blood,  
measured within 10 minutes

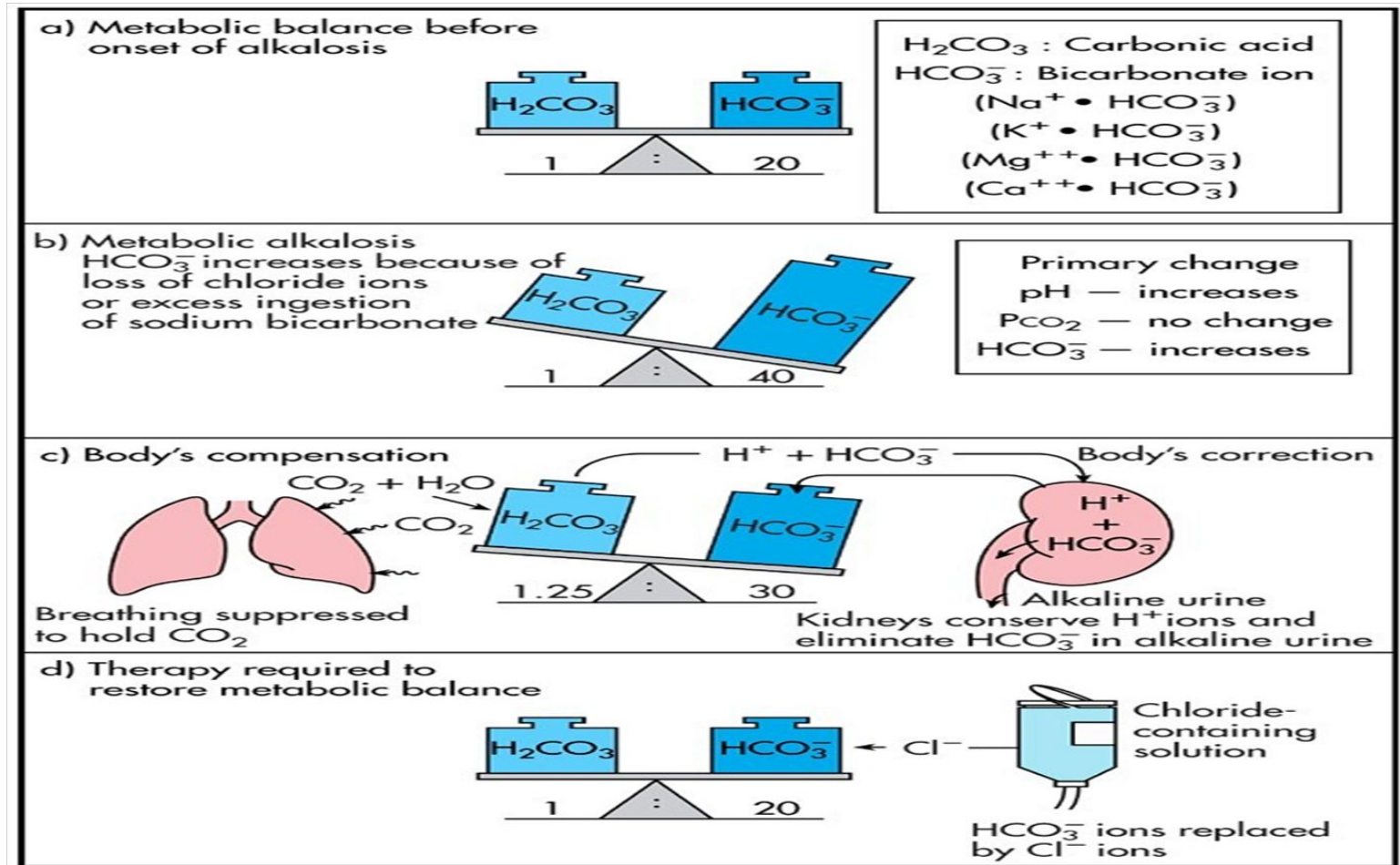
- Derived measures:

Bicarbonate (HCO<sub>3</sub><sup>-</sup>)

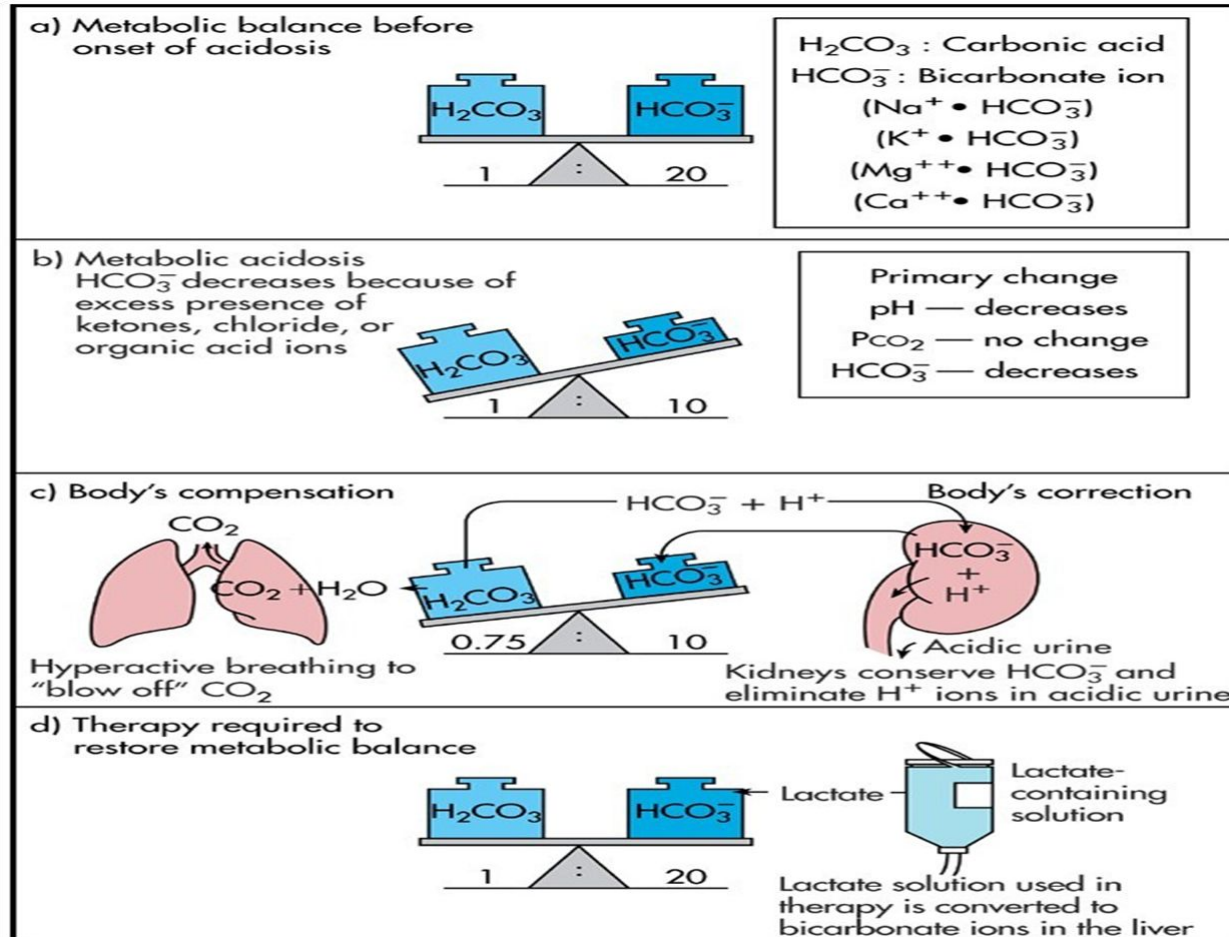
Normal Values:

pH = 7.35-7.45	(7.4)
HCO <sub>3</sub> <sup>-</sup> = 22 - 26 mEq / L	(24 mEq / L)
pCO <sub>2</sub> = 35 - 45 mm Hg	(40 mm Hg)

# Metabolic alkalosis

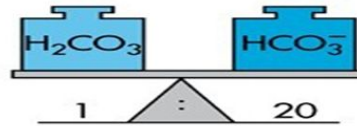


# Metabolic acidosis



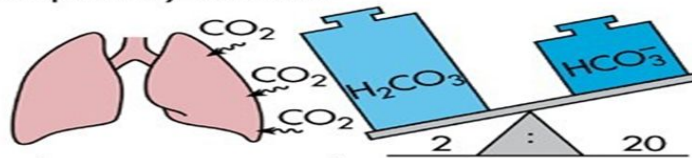
# Respiratory acidosis

a) Metabolic balance before onset of acidosis



$H_2CO_3$  : Carbonic acid  
 $HCO_3^-$  : Bicarbonate ion  
 ( $Na^+ \bullet HCO_3^-$ )  
 ( $K^+ \bullet HCO_3^-$ )  
 ( $Mg^{++} \bullet HCO_3^-$ )  
 ( $Ca^{++} \bullet HCO_3^-$ )

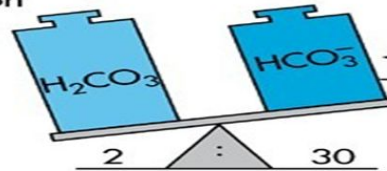
b) Respiratory acidosis



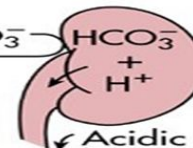
Breathing is suppressed, holding  $CO_2$  in body

Primary change  
 pH — decreases  
 $P_{CO_2}$  — increases  
 $HCO_3^-$  — no change

c) Body's compensation

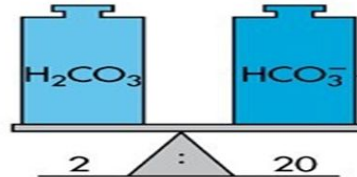


Body's correction  
 $H_2CO_3$



Kidneys conserve  $HCO_3^-$  ions and eliminate  $H^+$  ions in acidic urine

d) Therapy required to restore metabolic balance

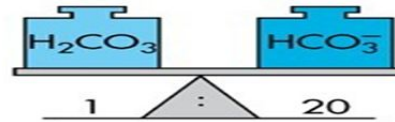


Lactate-containing solution

Lactate solution used in therapy is converted to bicarbonate ions in the liver

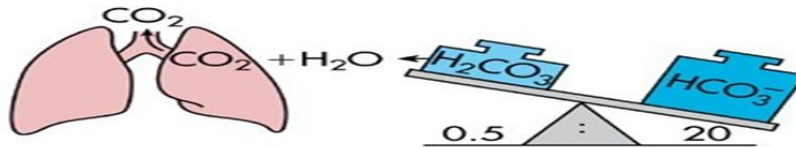
# Respiratory Alkalosis

a) Metabolic balance before onset of alkalosis



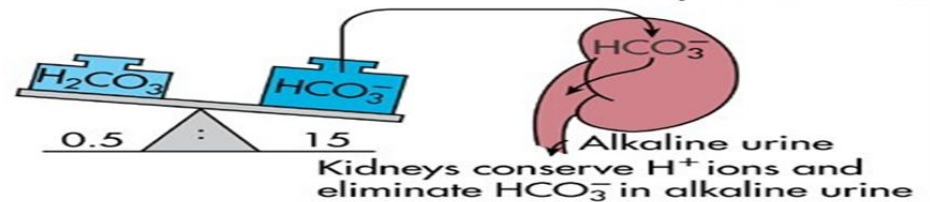
$\text{H}_2\text{CO}_3$  : Carbonic acid  
 $\text{HCO}_3^-$  : Bicarbonate ion  
 ( $\text{Na}^+ \bullet \text{HCO}_3^-$ )  
 ( $\text{K}^+ \bullet \text{HCO}_3^-$ )  
 ( $\text{Mg}^{++} \bullet \text{HCO}_3^-$ )  
 ( $\text{Ca}^{++} \bullet \text{HCO}_3^-$ )

b) Respiratory alkalosis

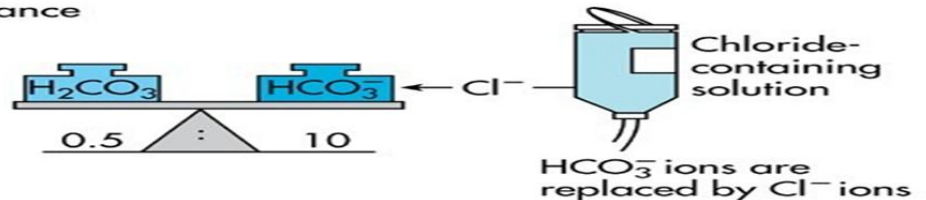


Primary change  
 pH — increases  
 $\text{PCO}_2$  — decreases  
 $\text{HCO}_3^-$  — no change

c) Body's compensation



d) Therapy required to restore metabolic balance





# Metabolic Acidosis

- pH 7.30
- PaCO<sub>2</sub> 40
- HCO<sub>3</sub> 15

# Metabolic Alkalosis


- pH 7.50
- PCO<sub>2</sub> 40
- HCO<sub>3</sub> 30


# Respiratory Acidosis

- pH 7.30
- PaCO<sub>2</sub> 60
- HCO<sub>3</sub> 26

# Respiratory Alkalosis

- pH 7.50
- PaCO<sub>2</sub> 25
- HCO<sub>3</sub> 23

- 
- What are the compensations?
  - • Respiratory acidosis --metabolic alkalosis
  - • Respiratory alkalosis --metabolic acidosis
  - • In respiratory conditions, therefore, the kidneys will attempt to compensate and visa versa.

- 
- Buffers kick in within minutes.
  - **Respiratory compensation** is rapid and starts **within minutes** and complete within **24 hours**. **Kidney compensation** takes **hours** and up to **5 days**

# Acid base disorder-worksheet

<u>Type of Disorder</u>	<u>pH</u>	<u>PaCO<sub>2</sub></u>	<u>[HCO<sub>3</sub>]</u>
Metabolic Acidosis	↓		↓
Metabolic Alkalosis			
Respiratory Acidosis			
Respiratory Alkalosis			

# Practice ABG's

pH 7.48	PaCO <sub>2</sub> 32	HCO <sub>3</sub> 24
pH 7.32	PaCO <sub>2</sub> 48	HCO <sub>3</sub> 25
pH 7.30	PaCO <sub>2</sub> 40	HCO <sub>3</sub> 18
pH 7.38	PaCO <sub>2</sub> 48	HCO <sub>3</sub> 28
pH 7.49	PaCO <sub>2</sub> 40	HCO <sub>3</sub> 30
pH 7.35	PaCO <sub>2</sub> 48	HCO <sub>3</sub> 27
pH 7.45	PaCO <sub>2</sub> 47	HCO <sub>3</sub> 29
pH 7.31	PaCO <sub>2</sub> 38	HCO <sub>3</sub> 15
pH 7.30	PaCO <sub>2</sub> 50	HCO <sub>3</sub> 24
pH 7.48	PaCO <sub>2</sub> 40	HCO <sub>3</sub> 30



# Answers:

1. Respiratory alkalosis
2. Respiratory acidosis
3. Metabolic acidosis
4. Compensated Respiratory acidosis
5. Metabolic alkalosis
6. Compensated Respiratory acidosis
7. Compensated Metabolic alkalosis
8. Metabolic acidosis
9. Respiratory acidosis
10. Metabolic alkalosis

# STEPS OF ASSESSING ABG

- *STEP 1:* Diagnose whether it is acidosis or alkalosis- (pH will help)
- *STEP 2:* Diagnose whether compensated or non compensated
- *STEP 3:* Diagnose whether it is metabolic or respiratory (Look at the value of bicarbonate and pCO<sub>2</sub>)

# Work sheet

- Diarrhea may lead to-----?
- Acid loss due to vomiting and gastric suction may lead to \_\_\_\_\_ alkalosis?
- Overuse of \_\_\_\_\_ may lead to metabolic alkalosis?

●

# Problem#1



- 67 year female known diabetic for past 20years presented with sudden onset of severe chest pain and Shortness of breath.
- ABG analysis showed:
- pH 7.36
- PCO<sub>2</sub> 33 mmHg
- HCO<sub>3</sub> 18 mmol/L
- Discuss the probable diagnosis.

## Problem #2

- A 30-year old man with DM presents with polyuria, polydipsia, fever, cough, and purulent sputum.
- His ABG shows the following Na<sup>+</sup> 140 / Cl<sup>-</sup> 104  
K<sup>+</sup> 7.0
- pH: 6.95
- pCO<sub>2</sub> : 33
- HCO<sub>3</sub> : 7.0
- Discuss the probable diagnosis.

# Problem#3

- 45 year old male was admitted to the emergency room with complaints of mild vomiting, associated with disorientation and muscular weakness. His blood investigations showed the following
  - pH =7.20      Na -137meq/l
  - HCO<sub>3</sub><sup>-</sup>=16mEq / L      Cl-108meq/l
  - pCO<sub>2</sub> = 34mm Hg      K -5.8
  - Glucose=685mg/dl
  - urea49mg/dl

# Problem #4

- 60 year male presents to the ED from a nursing home. You have no history other than he has been breathing rapidly and is less responsive than usual.

•  
•  
•  
•

- Na<sup>+</sup> 123 Cl<sup>-</sup> 99 HCO<sub>3</sub><sup>-</sup> 5
- pH 7.31 pCO<sub>2</sub> 10
- Discuss the probable diagnosis.

# Problem # 5

- 60year old man was admitted with severe abdominal pain, which started some 2 hours back.
- Clinically he was in a state of shock with distended abdomen. Femoral pulses could not be palpable
- His ABG shows the follows pH : 7.05
- pCO<sub>2</sub>: 26.3 mmHg
- HCO<sub>3</sub>: 7 mmol/L Discuss the probable diagnosis.





***THANK YOU  
FOR LISTENING!***